ON THE PHASE TRANSITIONS OF NASICON-TYPE Na₃Sc₂(PO₄)₃

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NASICON-type materials belong to the group of hetero-polyhedral framework compounds exhibiting ionic conductivity. These 'NAtrium Super Ionic CONductors' were first reported by GOODENOUGH et al. (1976) in consequence of a systematic exploration of cubic skeleton structures like pyrochlore and cristobalite-type carnegieite leading to the type material $Na_{1+x}Zr_2P_{3-x}Si_xO_{12}$. Nowadays, NASICONs are classified with the general formula $A_xM_2(XO_4)_3$ (A = Li, Na, K, Mg, Ca; M = Al, Sc, Y, Ti, Zr, V, Nb, Cr, Mn, Fe, Ga, In; X = Si, P, As, S). Due to their physical properties they gained interests as a solid electrolyte over decades, followed up by investigations as a potential insertion host for energy storage applications (JIAN et al., 2017).

The ionic conductivity is strongly symmetry correlated. Based on the pseudo-cubic, trigonal framework (ideally *R3c*) of the corner-linked MO₆ and XO₄ polyhedra three polymorphs are known for Na₃Sc₂(PO₄)₃, which exhibit up to now the highest sodium ion conductivity. The reversible states are (i) ionic insulator: monoclinic, low temperature α -phase below ~65 °C, (ii) ionic conductor: average rhombohedral, intermediate β -phase, and (iii) 'superionic' phase: rhombohedral, high temperature γ -phase above ~165 °C. All phase changes are induced by sodium ion order / disorder (COLLIN et al., 1986).

DSC measurements, Fig.1, show two distinct and smooth phase transitions ($\Delta T \sim 40$ °C) at ~10 °C ($\alpha \rightarrow \beta$) and ~160 °C ($\beta \rightarrow \gamma$) during heating. On cooling the $\beta \rightarrow \alpha$ transition is faint, moreover, the unsteady signal from 65 to -20 °C indicates a second β '-type phase. On dT-PXRD patterns, Fig. 2, this temperature range is characterized by the presence of sharp and diffuse incommensurate reflections. The $\beta \rightarrow \gamma$ transition is evident by a discontinuous increase of the *c*-axis and a small positive anomaly in the basal plane of the trigonal substructure. The general thermal expansion of Na₃Sc₂(PO₄)₃ is strongly anisotropic.



Figure 1. DSC curves of $Na_3Sc_2(PO_4)_3$, (-100 to 400 °C), heating (upper curve), cooling (lower curve).



Figure 2. dT-PXRD of Na₃Sc₂(PO₄)₃, (50-400 °C), $\beta \rightarrow \gamma$ phase transition (gray patterns).

COLLIN, G., COMES, R., BOILOT, J.P., COLOMBAN, PH. (1986): J. Phys. Chem. Solids, 47, 843-854. GOODENOUGH, J.B., HONG, H.Y.-P., KAFALAS, J.A. (1976): Mater. Res. Bull., 11, 203-220. JIAN, Z., HU, Y.-S., JI, X., CHEN, W. (2017): Adv. Mater., 29, 1601925.