MICROSTRUCTURE AND PHYSICAL PROPERTIES OF THERMALLY SPRAYED MULLITE BARRIER COATINGS

Kaindl, R.¹, Angerer, P.², Lackner, J.M.¹, Wießner, M.^{2,3} & Waldhauser, W.¹

¹JOANNEUM RESEARCH Forschungsgesellschaft mbH, MATERIALS – Institute for Surface Technologies and Photonics, Leobner Straße 94, 8712 Niklasdorf, Austria ²Materials Center Leoben Forschung GmbH, Roseggerstraße 12, 8700 Leoben, Austria ³Anton Paar GmbH, Anton Paar Straße 20, 8054 Graz, Austria reinhard.kaindl@joanneum.at

Chemistry, structure and thermal diffusivity of thermally sprayed mullite coatings with a thickness of about 0.5 mm were investigated by scanning electron microscopy with energydispersive X-ray spectroscopy, ex- and in-situ X-ray diffraction from room temperature up to 800 °C and by laser flash method up to 600 °C. The bulk composition of the coating is close to the pristine powder and to stoichiometric 3/2- (3Al₂O₃·2SiO₂) mullite. Individual splats are variable in their aluminum to silicon ratios and contain traces of calcium and iron. The XRD diffractogram of the as-deposited coating fitted best with a 3/2-mullite with the composition Al_{4.9}Si_{1.1} $O_{9.6}$ beside cubic γ -Al₂O₃ and a pronounced amorphous phase. In-situ hightemperature XRD analysis up to 800 °C showed the appearance of minor amounts of an unidentified crystalline phase at the highest temperature. The amorphous fraction displays almost no changes during the experiment. The thermally induced lattice expansion is around 2.6% for the lattice constants a and b and 7% for c, accompanied with a decreasing crystallite size from about 70 to 40 nm. The thermal conductivity λ of the coatings changes between 2.75 and 3.02 W (m K)⁻¹ in the temperature range 20 - 600 °C. This is significantly lower than for single crystal and dense mullite ceramics. Such extremely low values can be achieved by both increasing the porosity, number of microcracks and fraction of the amorphous phase and reducing the grain size and coating thickness.



Figure 1. Lattice parameters of the compositional series Al2(Al2+2xSi2-2x)O10x versus Al2O3 content.