

**SYNTHESIS, SYNCHROTRON DIFFRACTION STUDY AND TWINNING IN
Na₂Ca₄Mg₂Si₄O₁₅ – A HETEROPOLYHEDRAL FRAMEWORK COMPOUND**

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The formation of polycrystalline Na₂Ca₄Mg₂Si₄O₁₅ from solid state reactions has been studied between 800 and 1050 °C. Single crystals of the compound have been grown in a closed platinum capsule by slow cooling in the temperature range between 1300°C and 1000°C. Basic crystallographic data are as follows: monoclinic symmetry, space group *P12/c1*, $a = 7.1717(3)$ Å, $b = 5.3512(2)$ Å, $c = 16.4789(7)$ Å, $\beta = 90.911(4)^\circ$, $V = 632.33(4)$ Å³, $Z = 2$.

A conspicuous feature of the crystals is an intensive lamellar non-merohedral twinning clearly observable already under a petrographic microscope. The diffraction pattern can be explained as a superposition of two reciprocal lattices with a two-fold axis parallel to [001] being the twin element. Using synchrotron radiation (X06DA beamline at the Swiss Light Source, Paul Scherrer Institut, Villigen, Switzerland) it was possible to solve the crystal structure of Na₂Ca₄Mg₂Si₄O₁₅ from a twinned data set. Least-squares refinements resulted in a residual of $R(|F|) = 0.031$ for 2899 observed reflections with $I > 2\sigma(I)$ and 127 parameters. The crystal structure contains both [Si₂O₇]-dimers and insular [SiO₄]-moieties. Tetrahedra and [MgO₆]-octahedra form a three-dimensional framework whose topological characteristics have been studied. The remaining calcium and sodium cations are distributed among four crystallographically independent positions located in voids of the network.

On a microscopic scale the twinning observed in the diffraction experiments could be explained by the existence of 2₁-screw axis parallel to [001] in ($\frac{1}{4}, 0, z$) mapping both domains onto each other. A comparison with related compounds having an A⁺₂B²⁺₆Si₄O₁₅ stoichiometry is presented.

More than 25 years after its first observation in refractories (HAUSNER & SUPPNER, 1992) our investigation clarifies the crystal structure of a silicate that is of relevance for both Materials science and high pressure research (BINDI et al., 2015).

HAUSNER, R., SUPPNER, M. (1992): Powder Diffr., 7, 36-37.

BINDI, L., SAFONOV, O.G., ZEDGENIZOV, D.A. (2015): Contrib. Mineral. Petr., 170, 14, 1-11.