

Long-term changes in precipitation and stream water chemistry in small forest and moorland catchments at Beddgelert Forest, north Wales

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Abstract

Changes in the chemistry of bulk precipitation and stream water between 1982 and 2000 are described for small moorland and forest catchments located within Beddgelert Forest in north Wales. Two forest catchments were partially clearfelled in 1984 (D2; 68% and D4; 28%) whilst a third (D3) remained as an unfelled control until autumn / winter 1998/99 when partial felling took place in the headwaters. Over the monitoring period, the annual mean pH of bulk precipitation increased from 4.6 to 5.1 whilst the annual mean non-seasalt sulphate concentration decreased from 0.53 mg S l⁻¹ in 1985 to 0.24 mg S l⁻¹ in 2000. Since 1985, the annual wet deposition flux of non-seasalt sulphur decreased by 50% to 8.4 kg S ha⁻¹ yr⁻¹ in 2000. Annual mean inorganic nitrogen concentrations and annual wet deposition fluxes have remained relatively unchanged since 1982. The decrease in atmospheric sulphur deposition is reflected by decreased annual mean concentrations of non-seasalt sulphur, acidity, aluminium and calcium in all four streams irrespective of clearfelling activities. Annual variations in nitrate-N and potassium concentrations in the forest streams, largely determined by pulses of leaching following forest clearance, had no effect on stream acidity. In common with UK upland catchments, annual mean concentrations of dissolved organic carbon have increased from about 1 mg C l⁻¹ in 1985 to between 1.5 and 2 mg C l⁻¹ in 2000, although there is considerable year to year variability. Two boreholes drilled adjacent to catchments D3 and D4 have confirmed the presence of alkaline, base rich groundwater at Beddgelert. Although the boreholes are only 150 m apart, there are large differences in chemistry suggesting that different groundwater reservoirs have been intercepted providing further evidence of the complexity and heterogeneity of groundwater systems in upland catchments.

Keywords: acid deposition, acidification, recovery, forestry, clearfelling, trends, Beddgelert, streams, rainfall

Introduction

The Beddgelert forest clearfelling study began in 1982 in response to concerns about the potential environmental effects of widespread harvesting of mature conifer plantations in the UK uplands (Blackie *et al.*, 1980). These concerns focused around issues of soil nutrient depletion and acidification associated with biomass removal and leaching losses at harvest, plus the debate over the respective roles of acid deposition and forestry in surface water acidification (Harriman and Morrison, 1982; Stoner *et al.*, 1984). Whilst relevant information was available from studies in North America (e.g. Likens *et al.*, 1977; Bormann and Likens, 1979; Feller, 1977), the combination of site type

and forest management practice found in the UK was not represented. Many north American studies focused either on clearcutting of primary forest containing trees of mixed ages or on plantations established on clearcut primary forest in which case felling debris was often subject to controlled burning (Brown and Krygier, 1971). In contrast, UK forest practice in the early 1980s involved clear-felling entire stands of even-aged conifers, mainly Sitka spruce (*Picea sitchensis* (Bong.) Carr.), leaving perhaps a few trees or groups of trees for amenity purposes. Conventional harvesting techniques removed the stem only, leaving felling debris consisting of needles, branches and stems (< 7 cm diameter) on site. This debris provided a potential source of

nutrients for the succeeding crop, which was generally planted within two years.

The CEH Bangor study in Beddgelert forest was originally conceived as a replicated, block design, plot experiment comparing the biogeochemical effects of conventional harvesting with those of whole tree harvesting in which all the above ground biomass was removed. However, the location of the experiment provided an opportunity to monitor water quality responses to felling in three first order streams within the forest. A fourth stream adjacent to the forest, draining semi-natural moorland, was also sampled. As the catchments were located in an acid sensitive area (Hornung *et al.*, 1995) receiving large inputs of acid deposition (Reynolds *et al.*, 1999), the sites provide valuable information on the hydrochemical response to changing acid inputs over the last 20 years.

Much of the published work from Beddgelert has focused on the short to medium term dynamics of the major nutrients in response to harvesting, that is of the order of one to three years (e.g. Fahey *et al.*, 1991a,b; Stevens *et al.*, 1989, 1993, 1995). Particular focus was placed on nitrogen concentrations and fluxes (Stevens and Hornung, 1988, 1990; Stevens *et al.*, 1993) including some detailed process studies of the controls on nitrate leaching following harvesting (Emmett and Quarmby, 1991; Emmett *et al.*, 1991a,b). A broader consideration of the water quality impacts of felling in Beddgelert forest (covering a monitoring period of about seven years) was presented by Reynolds *et al.* (1995). The hydrochemistry of sites in Beddgelert forest was also described as part of a broader investigation into the interactions between atmospheric nitrogen deposition and tree age (Stevens *et al.*, 1993, 1995; Hughes *et al.*, 1994).

Although the soil plot studies have been discontinued for many years, monitoring of bulk precipitation and stream water chemistry has continued. This paper summarises these data to provide a long-term perspective on the earlier, more intensive studies and to show how the streams have responded over a twenty year period to changes in forest management (clearfelling) and acid deposition inputs.

Sites and methods

Beddgelert forest is located in the mountains of Snowdonia in north Wales. The forest study catchments D2 (1.4 ha), D3 (4.7 ha) and D4 (6.1 ha) are located in a north-east facing former glacial cwm known as Cwm Du and the moorland site (D6) is in the adjacent Cwm Marchnad. The altitude range of the catchments is between 250 and 700 m with an average annual rainfall of *c.*2800 mm (1982–2000 at 330 m altitude).

The geology of the two cwms is primarily Ordovician slates with small intrusions of dolerite and an area of micro-granite at the top of Cwm Marchnad. Glacial and slope deposits comprising locally derived clay-rich tills and compacted sandy gravels occur widely in Cwm Du whilst soils comprise a variety of acid upland types, dominated by stagnopodzols. The main area of forest in Cwm Du was established between 1931 and 1936 on agriculturally unimproved semi-natural grassland and heath. The trees were planted without ploughing as a 50:50 mixture of Sitka and Norway spruce, but subsequent thinning reduced the proportion of Norway spruce to 10%. In 1984, 62% of D2 and 28% of D4 were clearfelled, whilst D3 remained as an unfelled control site with closed canopy forest and no ground flora apart from scattered ferns and bryophytes. The top parts of D3 and D4 were clearfelled in the autumn and winter of 1998/99. The non-forested Cwm Marchnad comprises acid grassland dominated by *Nardus*, *Festuca* and *Molinia* and is grazed at low intensity by sheep.

SAMPLE COLLECTION

Water quality monitoring has been undertaken for rainfall, stream runoff and groundwater. The sampling programme was as follows. Rainfall was collected using a continuously open collector located in a clearing in Cwm Du. Samples were collected every two weeks from January 1982 until January 1992 when monthly sampling was introduced. Samples of stream water were collected weekly from D2, D3 and D4 from January 1982 until January 1992 when monthly sampling was introduced. Weekly sampling at D6 commenced in March 1983 and continued until the introduction of monthly sampling in January 1992. The data presented in this paper run to January 2001.

During the summer of 1994 two boreholes were drilled in Cwm Du immediately downslope of catchments D4 (hole BG1) and D3 (hole BG2). Initially, the boreholes were sampled every week between September 1994 and January 1995 and subsequently fortnightly until December 2000.

Within the study, hydrological measurements have also been taken. In the case of rainfall, the amount of rain falling during the sampling period was recorded from the bulk deposition collectors. A limited and incomplete record of water level at the time of sampling is available from small V-notch weirs installed in streams D3 and D4. These ceased operation in April 1989 for D4 and March 2000 for D3. For groundwater, the height of the water table was measured manually on each sampling occasion.

CHEMICAL ANALYSES

On return to the laboratory in Bangor, pH was determined

electrometrically on an unfiltered subsample. The remainder of the sample was filtered through a 0.45 µm membrane filter prior to analysis for major cations, anions, dissolved organic carbon (DOC) and silicon.

From October 1994 onwards, alkalinity has also been determined on the unfiltered sample by Gran titration. Since 1982, sodium and potassium have been determined by flame emission spectrometry while calcium, magnesium and manganese have been determined by flame atomic absorption spectrometry using a Perkin Elmer model 280 instrument. Up to September 1984, nitrate, chloride and ammonium were determined by continuous flow colorimetry after which ion chromatography was used. The methods were extensively cross-checked to ensure that no bias was introduced into the data. Sulphate was determined by a barium chloride turbidimetric method until the introduction of ion chromatography in September 1984. The barium chloride method was very insensitive at the concentration range encountered in the samples, so that data prior to September 1984 have been discarded. Since 1982, silicon has been determined by continuous flow colorimetry using the molybdenum blue method. DOC was determined from 1984 onwards using continuous flow colorimetry in which the samples were acidified and purged of CO₂ with nitrogen followed by UV digestion with potassium persulphate and determination of evolved CO₂ using weakly buffered phenolphthalein. Total monomeric aluminium has been determined since 1990 using the method proposed by Seip *et al.* (1984) followed by colorimetric analysis using pyrocatechol violet (Dougan and Wilson, 1974).

Results

BULK PRECIPITATION CHEMISTRY

Summary

The chemical composition of bulk precipitation at Beddgelert is summarised as annual rainfall-volume weighted means and ranges for the period 1982–2000 inclusive in Table 1. The excess or non-seasalt components have been calculated using theoretical seasalt ratios, assuming all sodium originated as seasalt. Bulk precipitation is typically dilute, (mean conductivity 22 µS cm⁻¹) and acidic (mean pH 4.76). The overall mean sodium to chloride ratio (0.56) is very close to the theoretical value for seawater of 0.55 and seasalts account for 78% of the total ion sum. Approximately 66% of the sulphur in precipitation originates from non-seasalt sources and of this about 10% is contributed by the oxidation of marine biogenic dimethylsulphide (McArdle *et al.*, 1998).

Table 1. Rainfall volume weighted mean, minimum and maximum solute concentrations in bulk precipitation collected at Beddgelert forest between 1982 and 2000 inclusive.

	Mean	Min	Max
pH	4.76	3.48	6.92
Na (mg l ⁻¹)	2.5	<0.1	12.2
K (mg l ⁻¹)	0.11	<0.01	0.85
Ca (mg l ⁻¹)	0.24	0.02	6.00
Mg (mg l ⁻¹)	0.31	<0.01	2.16
NH ₄ -N (mg l ⁻¹)	0.22	<0.01	3.46
NO ₃ -N (mg l ⁻¹)	0.18	<0.01	2.30
¹ SO ₄ -S (mg l ⁻¹)	0.59	0.15	4.50
¹ nmSO ₄ -S (mg l ⁻¹)	0.39	<0.01	4.11
Cl (mg l ⁻¹)	4.39	0.25	21.00
¹ DOC (mg l ⁻¹)	0.57	0.05	7.50

¹Data from 1985 to 2000 inclusive.

Changes in concentration and flux over time

Downward trends in excess sulphate concentration and acidity were reported for Beddgelert by Reynolds *et al.* (1999). Subsequently, these trends have been sustained (Fig. 1) such that the annual rainfall weighted mean non-seasalt sulphate-S concentration has more than halved from 0.53 mg S l⁻¹ in 1985 to 0.24 mg S l⁻¹ in 2000. Over the same period, pH has increased from 4.6 to 5.1, making the decrease in annual mean sulphate and hydrogen ion concentrations approximately equal (17 to 18 µEq l⁻¹). The annual mean equivalent ratio of excess sulphate to nitrate has also decreased from 2.2 in 1985 to 1.5 in 2000. Sulphate remains the dominant acid anion in bulk precipitation, but its importance relative to nitrate has declined. Annual mean concentrations of seasalt ions have varied in a broad cyclical pattern with maxima in the mid-1980s, 1990 and 1997 (Fig. 1). Excess calcium concentrations also show similar cyclical variations, with minima in 1986, 1991 and 1996.

Annual fluxes have been estimated for the site using precipitation volumes measured in the rain chemistry collector. Whilst these volumes may not be 'accurate' in absolute terms, since the collectors are not at ground level, they should provide a reliable indicator of inter-annual variability in rainfall totals.

There has been a clear decline in non-seasalt sulphur deposition since 1985 (Fig. 2). In 2000, sulphur deposition (8.4 kg S ha⁻¹ yr⁻¹) was approximately half that in 1985 (17.1 kg S ha⁻¹ yr⁻¹). The decline has not been linear and large fluctuations have occurred over the 16 year period with the lowest deposition fluxes recorded in 1995 (6.9 kg S ha⁻¹ yr⁻¹) and 1999 (6.5 kg S ha⁻¹ yr⁻¹). These fluctuations cannot be

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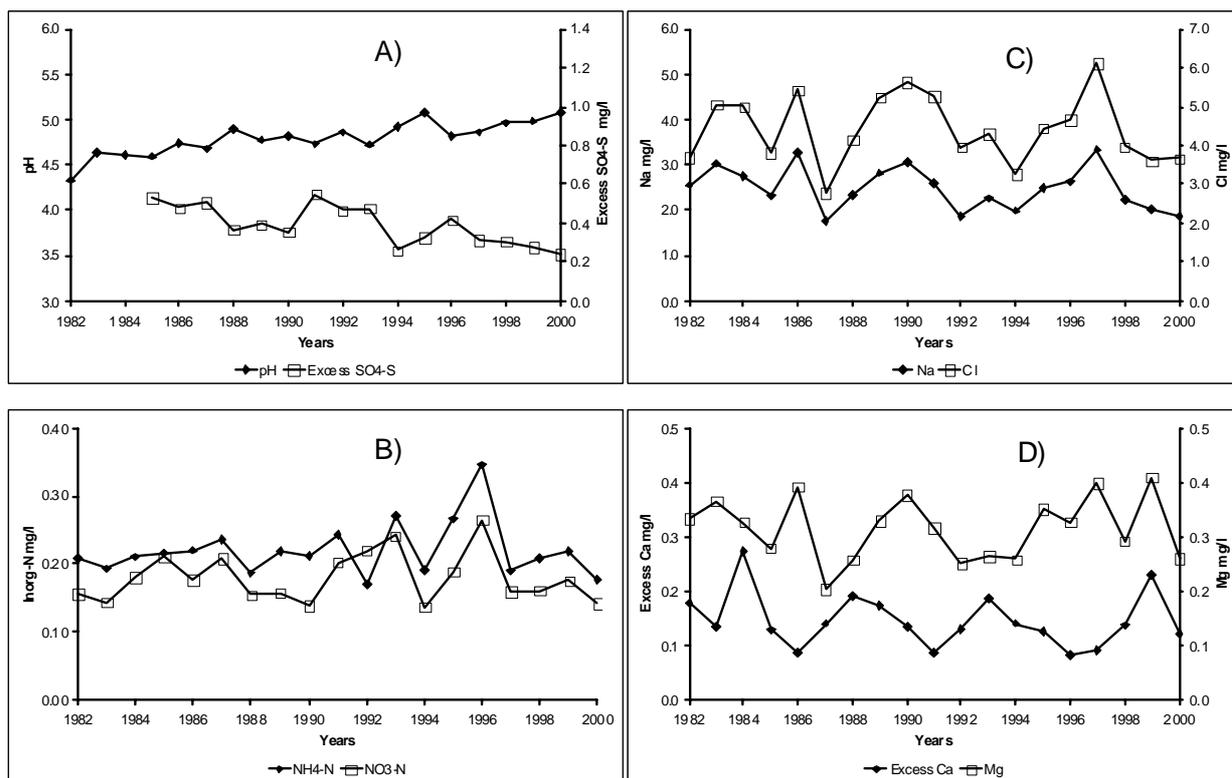


Fig. 1. Time series plots of (A) annual mean pH and annual mean non-seasalt sulphate-S concentrations; (B) annual mean inorganic nitrogen concentrations; (C) annual mean sodium and chloride concentrations; (D) annual mean non-seasalt calcium and magnesium concentrations in bulk precipitation at Beddgelert Forest.

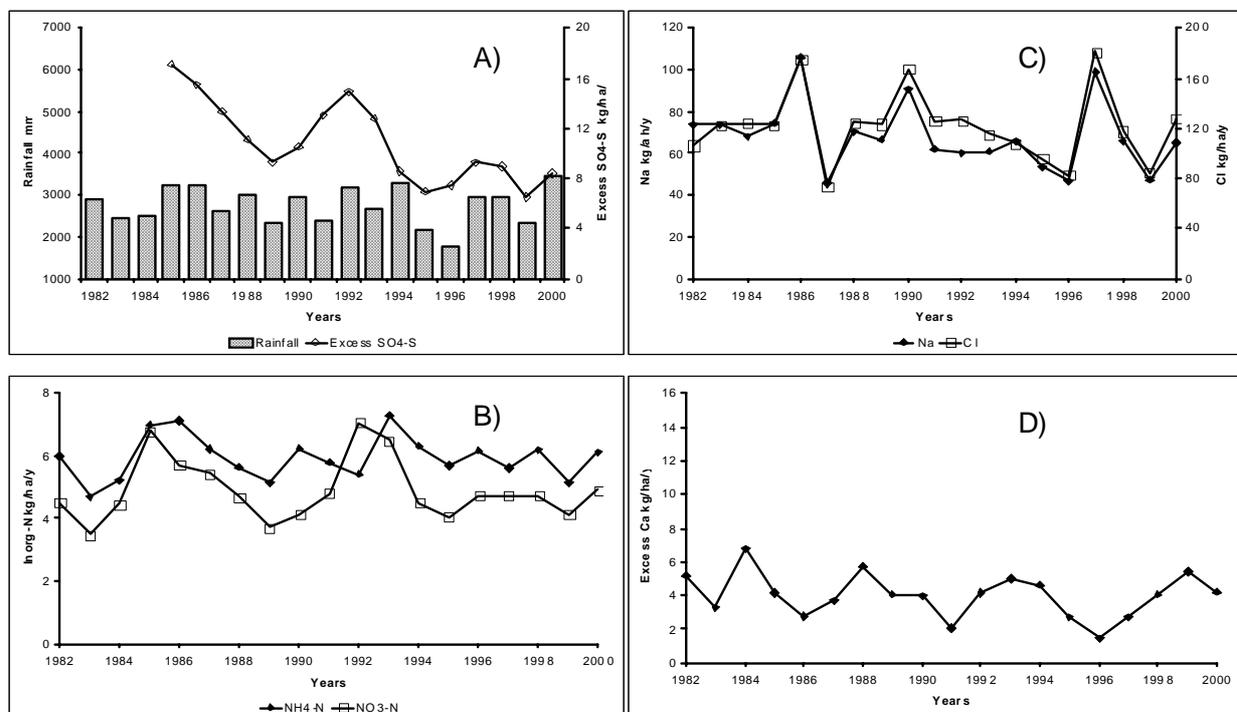


Fig. 2. Time series plots of (A) annual rainfall and non-seasalt sulphate-S deposition; (B) annual inorganic nitrogen deposition; (C) annual sodium and chloride deposition; (D) annual non-seasalt calcium deposition at Beddgelert forest.

attributed solely to variability in annual rainfall amounts, although these influence year to year differences in deposition. Annual deposition of inorganic nitrogen and seasalts show no clear trends although again there are large year to year fluctuations, which are not directly correlated with rainfall amount (Fig. 2). A weak downward trend in non-seasalt calcium deposition is discernible, but with large inter-annual variations in annual fluxes.

STREAM WATERS

Summary

The long-term average chemical composition of the three forest streams (D2, D3 and D4) is very similar (Table 2) despite the changes in forest cover as a result of clearfelling. All three streams are very acid with negative alkalinities. In comparison to D6, the forest streams are more acid, more

aluminium rich and contain much higher concentrations of seasalts and nitrate. For some solutes (acidity, aluminium, base cations, silicon and sulphate; Figs. 3 to 5), these broad differences have been maintained throughout much of the monitoring period. These differences probably reflect subtle differences in bedrock geology between the forest catchments and D6 and the continuing presence of significant areas of mature forest canopy. For the nutrients, a more complex pattern reflects transient pulses of nitrate and potassium in response to felling and longer-term changes in nutrient release as the new forest develops. Nitrate is the main form of inorganic nitrogen in the streams and mean concentrations of ammonium are close to the detection limit. The consistently high streamwater nitrate concentrations in the unfelled forest catchment (D3) prior to felling activity in 1998/9 have been attributed to the effects of stand age combined with large atmospheric inputs of inorganic

Table 2. Arithmetic mean, minimum and maximum in solute concentrations in streams draining forest and moorland catchments at Beddgelert forest.

	D2	D3	D4	D6
pH	4.62 3.95-5.61	4.66 4.04-5.36	4.63 4.05-5.44	4.92 4.36-5.55
Alk ($\mu\text{Eq l}^{-1}$)	-13.0 -62.3-+8.0	-18.5 -63.4-+1.7	-19.0 -77.7-+5.1	-10.8 -43.3-+3.4
Na (mg l^{-1})	6.3 3.1-12.4	6.7 3.1-16.5	5.5 2.4-12.3	3.7 2.2-7.5
K (mg l^{-1})	0.25 0.01-1.28	0.23 0.01-3.59	0.19 0.01-1.76	0.10 0.01-1.22
Ca (mg l^{-1})	1.02 0.34-3.12	1.94 0.21-4.10	1.33 0.13-3.90	0.78 0.42-1.60
Mg (mg l^{-1})	0.84 0.17-1.60	0.97 0.52-1.76	0.81 0.14-1.9	0.54 0.03-1.74
Mn (mg l^{-1})	0.16 0.02-0.98	0.21 0.02-1.87	0.17 0.01-0.92	0.05 0.01-3.00
Al (mg l^{-1})	0.43 0.04-1.20	0.50 0.07-1.16	0.53 0.04-1.32	0.11 0.01-0.60
NH ₄ -N (mg l^{-1})	0.05 <0.01-0.30	0.05 <0.01-0.68	0.05 <0.01-0.37	0.05 <0.01-0.68
NO ₃ -N (mg l^{-1})	0.52 0.01-2.00	0.78 0.18-3.55	0.75 0.12-3.46	0.12 0.01-1.20
SO ₄ -S (mg l^{-1})	1.89 0.43-5.7	2.27 1.16-4.10	1.91 0.86-4.20	1.17 0.48-3.00
Cl (mg l^{-1})	11.6 4.0-29.0	12.3 3.9-36.0	10.3 3.4-23.0	6.6 1.2-15.0
Si (mg l^{-1})	0.92 0.09-1.40	1.01 0.29-1.52	0.94 0.26-2.00	0.56 0.10-1.10
DOC (mg l^{-1})	1.42 0.08-18.0	1.21 0.08-9.30	1.03 0.08-11.00	1.01 0.08-6.30
Cond ($\mu\text{S cm}^{-1}$)	56 29-91	70 38-103	62 39-124	36 21-54

nitrogen (Stevens *et al.*, 1993, 1995; Emmett *et al.*, 1995). This catchment has been described as 'nitrogen saturated' (Stevens *et al.*, 1990), with seasonal maxima in stream water nitrate concentrations in summer; this is typical of 'nitrogen saturated' catchments (Stoddard, 1994) and the reverse of the more common pattern observed in the moorland catchment (Stevens *et al.*, 1993).

Changes in concentration over time

Different water quality determinands have shown several types of trend over time across the four catchments. The salient features are as follows.

- **Sulphur, acidity and aluminium** (Fig. 3). All four streams show a steady decline in annual mean sulphate-S concentrations since 1985 which broadly reflect the overall 50% decrease in excess sulphate-S deposition. The trends in the streams can be approximated by simple linear regression models (all significant at $p < 0.01$; r^2 values of between 0.6 and 0.8). The greatest rate of change in annual mean sulphate concentration occurred in D2 corresponding to a decline of 0.82 mg S l^{-1} in 16 years; equivalent to halving the annual mean concentration since 1985. A slower rate of change was observed in stream D6 corresponding to a decrease of 0.47 mg S l^{-1} equivalent to about 44% of the 1985 value. The downward trends are evident for excess sulphate-S (calculated assuming all chloride is of seasalt origin), but with more inter-annual variation due to fluctuations in stream water chloride concentrations. Since 1982, the pH in all streams has increased by about 0.5 units, although the forest streams have been consistently more acid throughout the monitoring period. In the moorland stream, the change in pH equates to a decrease in hydrogen ion concentration of about 50% from an annual mean value of $13.8 \text{ } \mu\text{Eq l}^{-1}$ in 1982. In the forest stream D2, for example, the decline in annual mean hydrogen ion concentration has been much larger; a 73% decrease from a 1982 value of $55 \text{ } \mu\text{Eq l}^{-1}$. In the moorland stream (D6), annual mean aluminium concentrations have declined steadily. The annual mean value in 2000 (0.06 mg l^{-1}) was one third of the 1990 value, whilst the corresponding decrease in the forest streams was about 60% from a 1990 value of 0.71 mg l^{-1} .
- **Nitrate and potassium** (Fig. 4). For the forest streams the inter-annual variations in mean concentrations of these nutrients reflect the responses to clear felling described previously (Stevens and Hornung, 1988; Reynolds *et al.*, 1995). The most striking changes are seen in stream D2 where annual mean nitrate-N concentrations peaked at 1.2 mg N l^{-1} in 1985 a year after felling, and dropped to values typical of the moorland stream (D6) by 1994. The response to more recent felling can be seen in streams D3 and D4. Superimposed on the felling response is a cyclical pattern of variation seen most clearly in D3 and D6, with concentration maxima in 1985, 1998 and 1996. These correspond to the patterns described for the UK Acid Waters Monitoring Network (AWMN) sites and at Plynlimon (Monteith *et al.*, 2000) which link long-term variations in stream and lake water nitrate concentrations to the North Atlantic Oscillation. This is likely to explain the sharp increase in nitrate concentrations in D2 between 1995 and 1996. No changes in forest management occurred during this period and it is unlikely that a change in forest nutrient cycling is responsible, as concentrations had returned to moorland values by 2000. For potassium the effects of felling in catchments D2 and D4 are evident from 1985 to 1987, and latterly in D3 and D4 between 1999 and 2000. Prior to 1995, potassium concentrations in both felled catchments (D2 and D4) had declined to values observed in D6. However, there was a sharp increase in potassium in D2 between 1995 and 1996, matching the increase in nitrate-N.
- **Seasalts (sodium, magnesium and chloride) and calcium** (Fig. 4 and 5). Sodium and chloride have no major sources within the catchment, so that patterns of variation are determined by atmospheric inputs which can in turn be linked to regional climatic influences such as the North Atlantic Oscillation (Evans *et al.*, 2001a). The large annual mean values in 1984 and 1990 reflect the influence of specific deposition events in the spring of those years. Data from daily rainfall collection at primary sites of the UK Acid Deposition Monitoring Network (ADMN) show that chloride deposition is highly episodic and primarily a winter phenomenon (NEG-TAP, 2001). However, there is considerable damping of the rainfall chloride signal in surface waters reflecting hydrological mixing and storage processes within the catchment (Neal and Kirchner, 2000) Although magnesium originates from both atmospheric and weathering sources, the former dominates and annual mean concentrations are highly correlated with sodium and chloride. In contrast, weathering is a major source of calcium and its inter-annual variation is determined by periods of low rainfall when stream chemistry is dominated by baseflow. However, in the forest streams, there is a weak downward trend in annual mean calcium concentration, which is not apparent in D6.

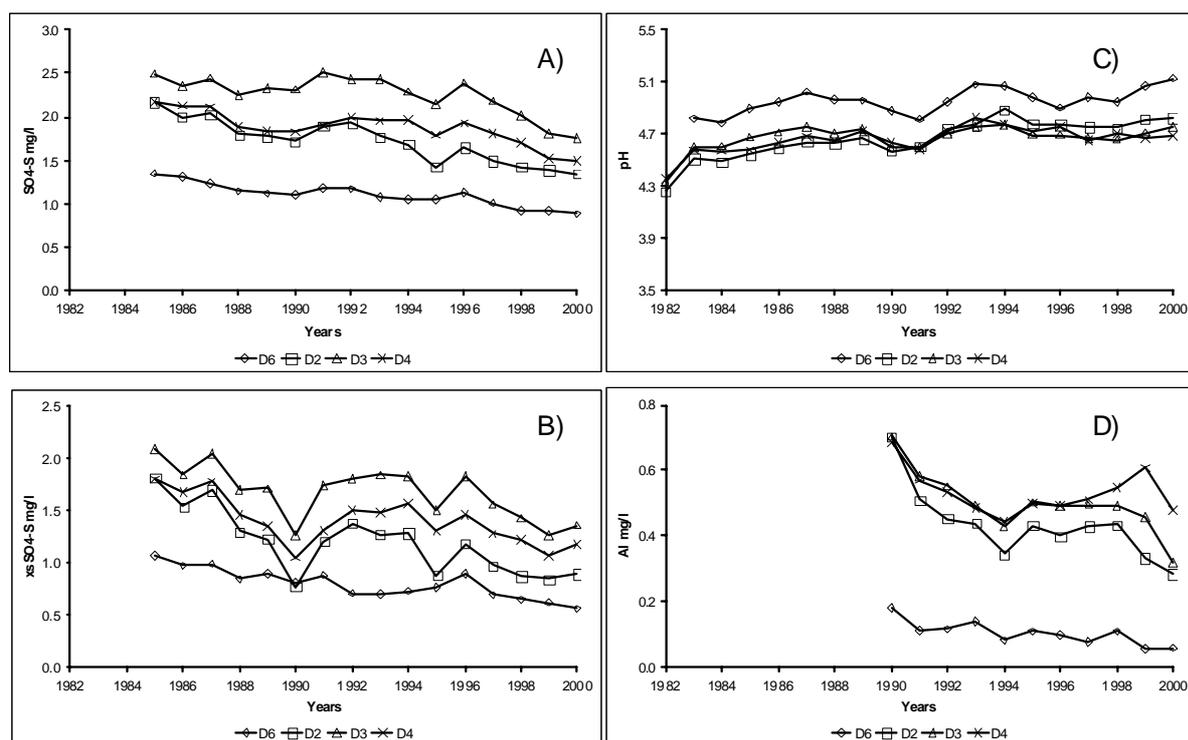


Fig. 3. Time series plots of (A) annual mean sulphate-S concentrations; (B) annual mean non-seasalt sulphate-S concentrations; (C) annual mean pH; (D) annual mean total monomeric aluminium concentrations in four streams at Beddgelert forest.

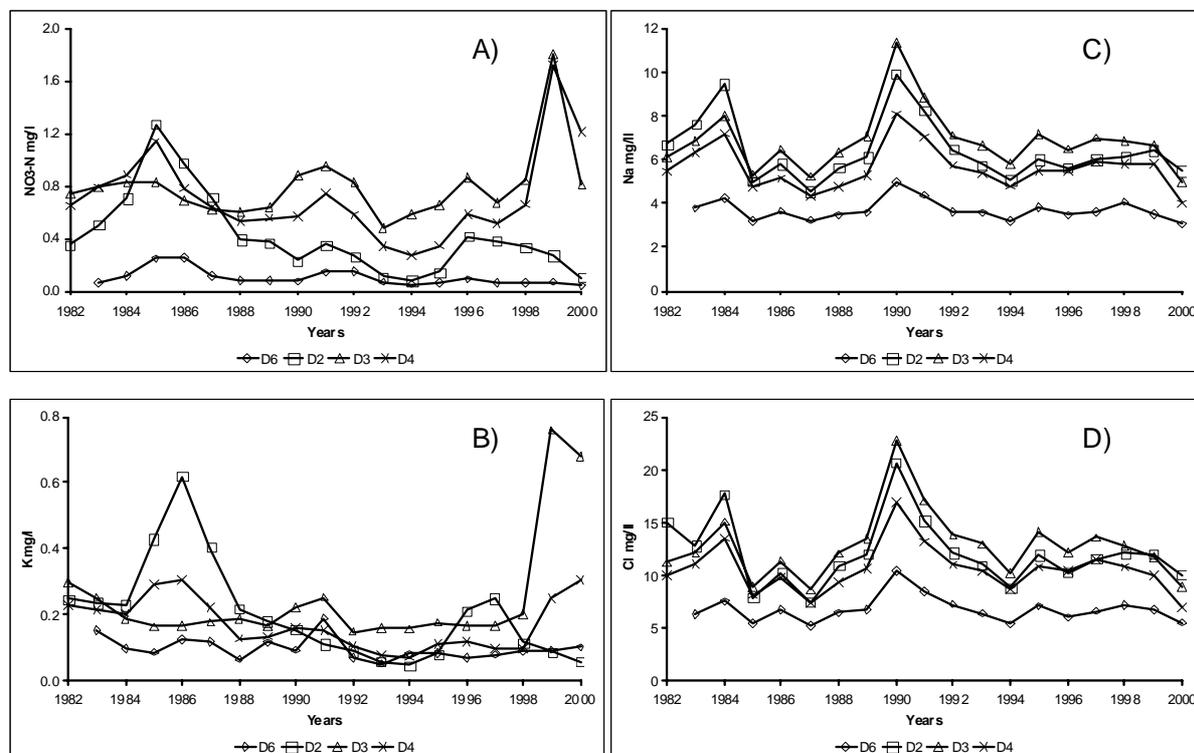


Fig. 4. Time series plots of annual mean concentrations of (A) nitrate-N; (B) potassium; (C) sodium; (D) chloride in four streams at Beddgelert forest.

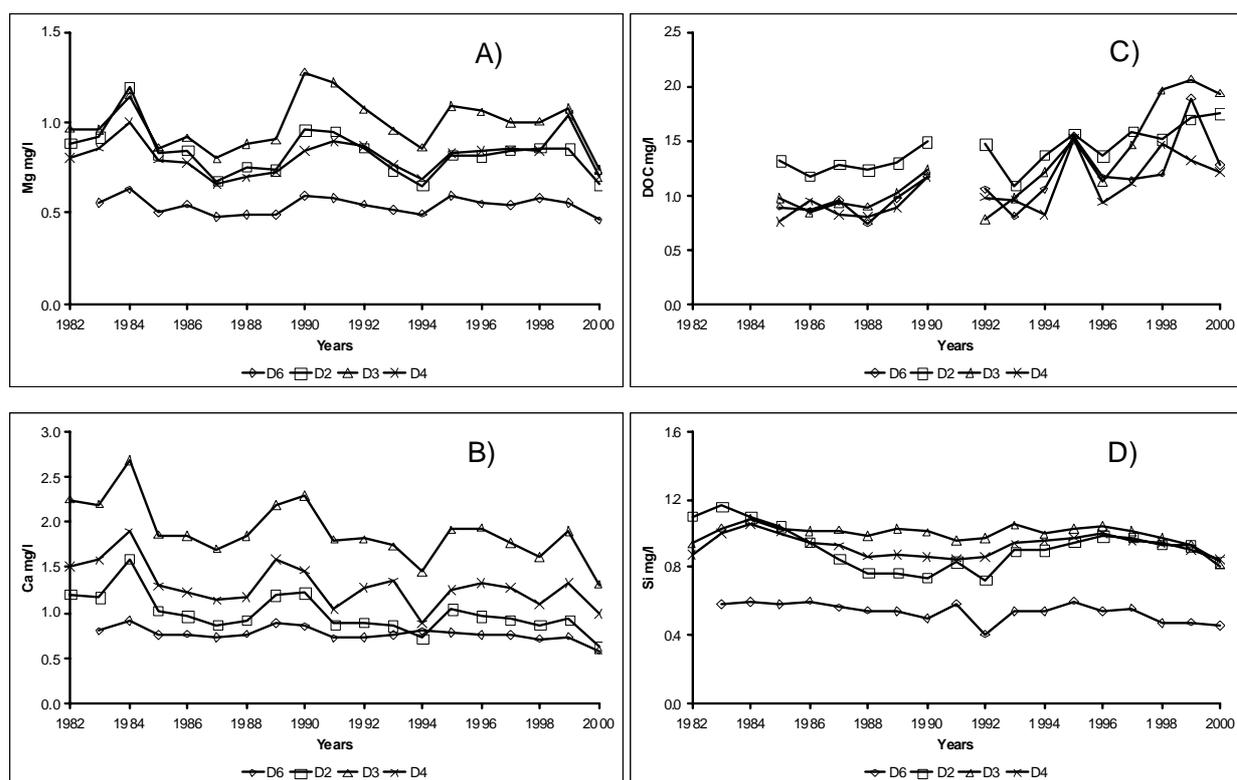


Fig. 5. Time series plots of annual mean concentrations of (A) magnesium; (B) calcium; (C) DOC; (D) silicon in four streams at Beddgelert forest.

DOC and silicon (Fig. 5). In common with many upland streams (Freeman *et al.*, 2001), DOC concentrations have increased in all four streams at Beddgelert, although there is considerable inter-annual variability. Annual mean concentrations of silicon are remarkably constant over time; in the forest streams between 0.9 and 1.1 mg l⁻¹ whilst D6 ranges between 0.4 and 0.6 mg l⁻¹. Silicon is derived primarily from weathering and variations in bedrock geochemistry may explain the differences between the moorland and forest streams. In all four streams, waters are just undersaturated with respect to quartz (Neal *et al.*, 2004c). The degree of undersaturation increases at high flows when water is derived from organic-rich soil horizons, which are depleted in silicate minerals.

GROUNDWATER

The chemistry of all four streams follows the classic response to changes in flow, becoming more acidic and aluminium-rich under high flows when soil water inputs dominate and more alkaline, calcium-rich at baseflow (Reynolds, *et al.*, 1986). To explain these fluctuations, a source of base rich water was required within the catchments. This could not be provided from the known chemistry of soil waters, which

are acid, aluminium-rich and depleted in calcium throughout the profile (Stevens *et al.*, 1989). Following the discovery of groundwater at Plynlimon (Neal *et al.*, 1997), two boreholes were drilled in Beddgelert forest in 1994; both encountered groundwater.

The chemical data from the boreholes (Table 3) contrast markedly with those from the streams (Table 2). The borehole waters are circumneutral, with high, positive alkalinities. Levels of dissolved carbon dioxide in these waters can be estimated using the approximation developed by Neal *et al.* (1998) where:

$$\text{EpCO}_2 \text{ H} \cdot (\text{Gran Alkalinity} + 10^{6-\text{pH}}) \cdot 10^{6-\text{pH}} / 6. \quad (1)$$

EpCO₂ is the degree of CO₂ saturation with reference to the atmosphere, Gran alkalinity has units of μEq l⁻¹ and the 10^{6-pH} term is an approximation of the hydrogen ion concentration in the same units. From this, BG1 and BG2 are respectively 18 and 8 times oversaturated with respect to atmospheric carbon dioxide. This means that on degassing the pH of these waters will increase to approximately 7.7 and 8.2 assuming surface waters are about 2 times over-saturated with respect to equilibrium with water in contact with the atmosphere.

Table 3. Arithmetic mean, minimum and maximum in solute concentrations in groundwater in forest catchments at Beddgelert forest.

	BG1	BG2
pH	6.73 5.86-7.23	7.49 6.73-7.85
Alk ($\mu\text{Eq l}^{-1}$)	712 286-1004	2291 1984-2579
EpCO ₂	18 8-117	12 5-68
Na (mg l^{-1})	4.4 3.7-5.3	23.3 14.3-25.4
K (mg l^{-1})	0.41 0.24-0.59	4.12 2.78-4.53
Ca (mg l^{-1})	15.7 7.9-22.0	30.6 20.4-33.3
Mg (mg l^{-1})	2.56 1.06-3.74	2.17 1.64-2.52
Mn (mg l^{-1})	0.42 0.01-1.68	0.17 0.01-1.53
Al (mg l^{-1})	0.01 <0.01-0.06	0.02 0.01-0.10
NH ₄ -N (mg l^{-1})	0.03 <0.01-0.41	0.02 <0.01-0.16
NO ₃ -N (mg l^{-1})	0.07 0.01-1.25	0.10 0.01-0.88
SO ₄ -S (mg l^{-1})	3.46 2.38-4.18	3.31 2.98-3.67
Cl (mg l^{-1})	9.59 7.92-11.35	11.15 9.93-12.52
Si (mg l^{-1})	3.34 2.01-4.22	3.05 1.73-3.72
DOC (mg l^{-1})	1.4 0.1-13.5	2.78 0.01-23.8
Cond ($\mu\text{S cm}^{-1}$)	122 78-154	266 163-291
Water level (m)	0.97 0.02-1.65	1.27 1.04-1.42

With the exception of sodium in BG1, the borehole waters are enriched in base cations, silicon, chloride and sulphate compared to the streams but depleted in aluminium. These waters are unlike any of the soil waters sampled at the site which are consistently acidic (pH 3.8–4.5), aluminium bearing (0.4–1.5 mg l^{-1}) and depleted in calcium (0.7–1.2 mg l^{-1}) throughout the profile (Stevens *et al.*, 1989).

Although the boreholes are located only 150 metres apart, there are notable differences in their hydrological and hydrochemical characteristics. The average depth of water in the boreholes was approximately one metre, but water level fluctuations in BG1 were more dynamic than in BG2.

Concentrations of calcium, potassium, sodium and chloride are much greater in BG2 which is less acid than BG1 with an average Gran alkalinity in excess of 2000 $\mu\text{Eq l}^{-1}$. This suggests that the two boreholes are intercepting separate groundwater reservoirs exposed to different weathering regimes; this is consistent with the high degree of spatial heterogeneity observed in groundwaters at Plynlimon (Hill and Neal, 1997; Neal *et al.*, 1997).

Discussion

The monitoring at Beddgelert covers a period during which UK annual SO₂ emissions declined by more than two thirds from 3720 kT SO₂ in 1985 to 1156 kT SO₂ in 2000 (NAEI, 2003). Correspondingly, annual wet deposition of non-seasalt sulphur at Beddgelert has approximately halved, with an annual rate of change very similar to that observed at UKADMN sites in northern England, central Wales and central southern England (NEG-TAP, 2001). The results at Beddgelert are consistent with long-term records at Plynlimon in mid-Wales where a much smaller relative decrease in non-seasalt sulphur concentration in bulk precipitation was observed over 17 years from 1983 (Neal *et al.*, 2001). Both studies provide examples of the spatially variable relationship between pollutant emission and deposition in the UK referred to as ‘non-linearity’ (NEG-TAP, 2001) in which the reduction in wet sulphur deposition is smallest at remote westerly sites but largest near the main emission sources. Sulphur emissions from shipping in the Atlantic, background contributions from North American sources and increases over time in the rates of dry deposition of SO₂, due to changes in the relative amounts of SO₂ and ammonia in the air, are thought to be responsible for the observed non-linearity (NEG-TAP, 2001).

Over the monitoring period, UK annual NO_x emissions have declined from 2540 kT in 1985 to 1512 kT in 2000, representing a 40% reduction (NAEI, 2003). However, there has been no discernible change in annual wet deposition of nitrate-N at Beddgelert; this is consistent with observations at Plynlimon and other western UK ADMN sites (NEG-TAP, 2001; Neal *et al.*, 2001) and more widely across many sites in Europe (Wright *et al.*, 2001). Ammonia emissions are very uncertain, but published data indicate a 13% decline from 341 kT in 1990 to 297 kT in 2000 (NAEI, 2001). However, between 1992 and 1997, annual emissions stabilised at about 324 kT. The lack of a trend in ammonium concentrations and wet deposition is therefore unsurprising and consistent with data at Plynlimon and across the UK ADMN (NEG-TAP, 2001; Neal *et al.*, 2001).

The decline in non-seasalt sulphur deposition at Beddgelert is reflected in the declining concentrations of

non-seasalt sulphate in both forest and moorland streams. Annual mean non-seasalt sulphur concentrations in streams D3 and D4, in which limited felling occurred prior to 1999, follow very similar, parallel trends. At Plynlimon, it is more difficult to discern a clear trend from amongst the inter-annual fluctuations in concentration (Neal *et al.*, 2001). The different responses at the two sites may be related to scaling, in that the Beddgelert catchments are smaller by a factor of 50 or more compared to those at Plynlimon. Clearfelling of nearly 70% of D2 appears to have caused a more rapid decline in sulphate concentrations, as removal of the majority of the forest canopy will have increased the water flux and decreased capture of cloud droplets and aerosols.

In conjunction with declining non-seasalt sulphur concentrations, stream acidity, aluminium and calcium concentrations have also decreased in the forest control stream (D3) in a manner consistent with observations at forested and non-forest acid sensitive surface water monitoring sites throughout the UK (Evans and Monteith, 2001; Harriman *et al.*, 2001), Europe (Evans *et al.*, 2001b; Skelkvåle *et al.*, 2003) and north America (Stoddard *et al.*, 1999). As with sulphate, there is an indication that large scale clearfelling of D2 has led to more rapid decreases in both calcium and aluminium compared to the other forest streams (prior to 1999), whereas there is no clear difference in pH trends amongst the forest streams. Reporting on acidity trends in the UK AWMN, Evans and Monteith (2001) suggested that trends in acidity-related variables were generally weaker in forest than in moorland systems. This does not seem to be the case at Beddgelert where stronger trends in calcium and aluminium are apparent in the forest streams. However, Evans and Monteith (2001) note that with only a small number of forest sites in the UK AWMN and considerable inter-site heterogeneity, it is difficult to generalise the effects of forestry on recovery from acidification. In a study of Scottish forest catchments, felling appeared to accelerate recovery trends with respect to acidifying components although forest streams were still more acid and contained higher concentrations of labile aluminium compared to moorland catchments (Harriman *et al.*, 2003). At Beddgelert, the clear geochemical differences between D6 and the forest streams and the effects of clear felling are confounding factors in a forest-moorland comparison of acidity trends.

Variations in nitrate and potassium in the forest streams are dominated by the effects of forest harvesting and climatic variability. Despite the large variations in nitrate-N in D2, no clear acidification effect associated with forest harvesting is discernible from either weekly time series pH data (Stevens *et al.*, 2001) or the annual mean values. The data record for stream water aluminium concentrations, however,

post-date the clearfelling nitrate pulse. This is unfortunate, since it is possible that aluminium leaching may have responded strongly to the nitrate pulse in these very acid streams as shown by earlier work across 67 first-order stream sites in Wales (Neal *et al.*, 1998). At Plynlimon, surface water acidification clearly accompanied the pulse of nitrate released at felling in the 13.7 ha, first order south2Hore catchment (Neal *et al.*, 2003). As the nitrate pulse declined, acidification was reversed such that after five years, the stream had a higher pH and lower aluminium concentrations than before felling. Complete felling of small, first order streams represents the end-member case study where the most extreme responses are likely to be seen. Even at this scale, a spectrum of responses reflect the highly heterogeneous nature of upland systems with respect to soil types, chemical reactivity and hydrological pathways (Neal *et al.*, 1998, 2004a, b). With an increase in scale to second and third order catchments, the net acidification effect of clearfelling may be hard to discern against other variations in water quality. This occurs when harvesting is phased across the catchment so that the effects of the standing forest, felling disturbance and post-felling recovery are integrated at the outflow (Neal and Reynolds, 1998; Neal *et al.*, 2004). In such cases, factors such as climatic variability and long-term changes in pollutant deposition can dominate variations in water quality (Neal *et al.*, 2004a, b and c)

In common with many upland catchments across the UK (Evans and Monteith, 2001), DOC concentrations have increased in both the moorland and forest streams over the last 20 years. This may be associated with increasing mobilisation of organic acids in response to a decrease in mineral acid inputs (Krug and Frink, 1983; Tipping and Hurley, 1988). However, it has been suggested that climatic controls are more important (Freeman *et al.*, 2001), given that increases in DOC are observed at sites which have received relatively little acid deposition (Worrall *et al.*, in press). Even so, the strength of the trend can be quite variable, even within adjacent catchments subject to similar environmental conditions (Neal *et al.*, 2001). The combination of increased pH and DOC in the streams at Beddgelert may have changed the speciation of aluminium which cannot be identified from the measurements of total monomeric aluminium. A decrease in the proportion of labile aluminium was reported at 10 out of 22 UK AWMN sites (Monteith and Evans, 2001). This reflects increased complexation by dissolved organic matter (Brown and Driscoll, 1993). This aluminium binding may well have resulted in a reduction in harm to aquatic biota as assessed using data from earlier toxicological studies (Baker and Schofield, 1982).

The occurrence of base-rich alkaline groundwater has

been confirmed at Beddgelert in common with upland acid-sensitive sites elsewhere in Wales (Neal *et al.*, 1997; 2003) and in Scotland (Soulsby *et al.*, 1998). The presence of groundwater helps to account for fluctuations in the calcium content of stream water in response to flow and the relatively large mass flux of calcium at the catchment outflows which could not be explained by sources within the soil profile at Beddgelert (Stevens *et al.*, 1989). The contrasting hydrochemistry and hydrology of the two boreholes further demonstrates the complexity and heterogeneity of groundwater systems in upland catchments (Neal *et al.*, 1997).

Conclusions

In response to declining industrial sulphur emissions, rainfall acidity and wet deposition of non-seasalt sulphur have decreased steadily since the early 1980s at Beddgelert Forest but the concentration or wet deposition of inorganic nitrogen is unchanged. There is a weak downward trend in non-seasalt calcium deposition, which may offset benefits following the decrease in sulphur deposition. Sulphate is still the dominant acid anion in precipitation, but its importance relative to nitrate has declined.

The long-term stream chemistry data from Beddgelert Forest provide evidence of recovery from acidification in response to decreasing wet sulphur deposition for a very acid-sensitive part of western Britain. The recovery trends in acidity related variables are evident in catchments under moorland, mature spruce forest and clear-felled / re-stocked forest. Forest nutrient cycling and harvesting have a major influence on stream water nitrate concentrations. A relatively short-lived nitrate leaching pulse follows harvesting. Subsequently, stream nitrate concentrations in the harvested catchment have approached values typical of the adjacent moorland stream, presumably due to forest re-growth.

Two boreholes drilled adjacent to the forest catchments have revealed shallow groundwater which is alkaline and base-rich in contrast to soil waters in the catchment which are acidic, base-poor and aluminium-rich. The groundwater provides a source of calcium which accounts for relatively high stream water calcium concentrations observed at baseflow and the relatively large mass flux of calcium at the outflows to the catchments calculated in earlier studies (Stevens *et al.*, 1989).

In terms of broader issues of forest management and surface water quality, the Beddgelert studies provide examples of an 'end member' water quality response to clear felling consistent with other work on small, first order catchments in the UK (Adamson and Hornung 1990; Reynolds *et al.*, 1995; Neal *et al.*, 1998, 2003). The clear

signals generated by such studies are invaluable for developing process understanding but they cannot necessarily be extrapolated directly to larger scale catchments, where the signals from a mosaic of forest management and other land uses are integrated at the catchment outflow. In such cases, the effects of climatic variability and long term changes in pollutant deposition may dominate patterns of variation in water quality. In regard to the latter, the long-term data sets at Beddgelert and other similar sites are of immense value in providing ground truth for model predictions and quantitative evidence of the effectiveness of emission control protocols for sulphur and nitrogen pollutants.

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