

Box and Gaussian plume models of the exhaust composition evolution of subsonic transport aircraft in- and out of the flight corridor

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Abstract. A box and a Gaussian plume model including gas-phase photochemistry and with plume dispersion parameters estimated from the few available plume observations are proposed and used for evaluation of photochemical transformations of exhausts from a single subsonic transport aircraft. The effects of concentration inhomogeneities in the plume cross section on the photochemical sources and sinks in the plume are analyzed for various groups of compounds. The influence of these inhomogeneities on the rate and on the mass of ambient air entrainment into the plume are studied also by comparing the box and the Gaussian plume model simulations during the first hours of their “life”. Due to the entrance of HO_x and NO_x from ambient air into the plume with rates varying from the wind shear and turbulence conditions, the rate of emitted NO_x oxidation in the plume is dependent on these and also on the background concentration levels of HO_x and NO_x .

1 Introduction

In the extensively studied sub- and supersonic aircraft exhaust atmospheric effects the problem of exhaust plume dispersion and photochemical transformations in the plume during the first minutes and hours of the plume “life” has been relatively less investigated (see Schumann, 1994; relevant papers in Schumann and Würzel (Eds.) 1994; WMO, 1995). Some model calculations, compared with very few measurements in the exhaust plumes of transport aircraft, suggest the possibly considerable oxidation of emitted NO_x into the HNO_3 and HO_2NO_2 by the OH and HO_2 contained in the ambient air, entrained into the plume during the 10 hours or so of plume “life”, when its gas and aerosol composition may be distin-

guished from the background air (Veenstra and Beck, 1994; Karol *et al.*, 1994). This phenomenon indicates that some significant corrections have to be made in the identifying sources and in the model estimations of large-scale and global effects of aircraft exhausts on the atmospheric composition and climate, where it has been assumed that emitted NO_x enters unchanged into the large-scale and global atmospheric systems (Ko and Douglass, 1993; Schumann, 1994; WMO, 1995). The preliminary results of model simulations reveal the obvious dependence of NO_x oxidation rate on the HO_x content in the ambient air and on the rate of its entrainment into the plume (Karol *et al.*, 1994). These results, obtained by using the simplified box model of the plume, need to be verified by the more accurate 2-D diffusion – dispersion plume model and for different ambient air compositions, which may occur in various flight conditions: in the air transport flight corridor and outside of it, in particular.

This work presents some results of subsonic aircraft exhaust plume composition evolution in its atmospheric dispersion regime as simulated by the 2-D diffusion model of plume dispersion. These are compared to the box model results with the same dispersion parameters, obtained from actual aircraft plume crossings and measurements (see Schlager *et al.*, 1994; Arnold *et al.*, 1994; Schulte and Schlager, 1994; Zheng *et al.*, 1994; Fahey *et al.*, 1995). The time evolutions of the inputs from the photochemistry in the plume and from the ambient air entrainment to the rate of gas content change in the plume cross section are calculated and compared, for box and for diffusion plume models with various dispersion parameters, and for various NO_x , NO_y and HO_x concentrations in the ambient air. The computed air composition evolution in the plume is compared with some observational data from the plume crossings mentioned.

2 Plume dispersion models and their parameters

The gas-phase photochemical processes are simulated in all the known jet, vortex and wake-dispersion regimes of

aircraft exhaust plume evolution, which occur in 1–10; 10–100 and more than 100 s of time correspondingly after leaving the engines. According to model results and observations (Miake-Lye *et al.*, 1993; Schumann, 1994) the conservation and some increase of plume cross section dimensions and its isolation from the ambient air are assumed during the jet and vortex regimes of the plume in various calculated variants. Plume dispersion occurs in the third regime, where the horizontal plume axis is adopted as the plume axis descends by not more than 200–300 m from the emission level due to dynamical and radiation causes (Miake-Lye *et al.*, 1993; Schumann, 1994). The intensive and complicated air turnover in the plume during the jet and vortex periods has lead to the adoption of the box regime for the photochemical processes there. Real deviations of gas concentrations from their homogeneity in the initial plume cross section for the dispersion regime may influence only slightly the plume photochemistry during its first few minutes.

The mass balance equation of the model is (Karol *et al.*, 1992):

$$\partial q/\partial t - Aq = f(q, t) \quad (1)$$

for the gas volume concentration $q(y, z, t)$ in the plume cross section with t , time; z , vertical; y , horizontal coordinate perpendicular to the plume axis; and $f(q, t)$, photochemical source – sink intensity. The air transport operator Aq for the box model has the form:

$$Aq = (q_e - q)dS/Sdt \quad (2)$$

where $S(t)$ is the plume cross section area; and q_e , the external gas concentration in the ambient air. For plume diffusion-dispersion where wind velocity field changes with the height according to (Karol *et al.*, 1992; Schumann and Konopka, 1994) the Gaussian distribution parameters in the y, z plane are:

$$\begin{aligned} \sigma_y^2(t) &= \sigma_{oy}^2 + 2K_y t + s_z^2 \sigma_{oy}^2 t^2 + 2s_z^2 K_z t^3/3 \\ \sigma_z^2(t) &= \sigma_{oz}^2 + 2K_z t; \quad \sigma_{yz}^2(t) = s_z t \cdot (\sigma_{oz}^2 + K_z t), \end{aligned} \quad (3)$$

where σ_{oy} , σ_{oz} are their initial values; K_y and K_z are the semi-empiric turbulent diffusion coefficients and s_z is the wind shear parameter.

As discussed in Karol *et al.*, (1994) the plume cross section ellipse area $S(t)$ is determined by its half-axes $2\sigma_y$; $2\sigma_z$ and it contains 91.1% of mass in the plume cross

section, distributed along the 2D Gaussian distribution. In the jet and vortex regimes of the plume, during its first 150 s, the increase of plume box dimensions from $4\sigma_z = 60$ m, $4\sigma_y = 100$ m to $4\sigma_z = 70$ m, $4\sigma_y = 400$ m is assumed, as determined by jets and vortex from the four engined Boeing-747 aircraft (Schumann, 1994; Schumann and Konopka, 1994). Then the dispersion regime begins, and its parameters are estimated from the actual plume and turbulence measurements (Schlager *et al.*, 1994; Kärcher, 1994; Schumann *et al.*, 1995).

These were carried out in plume crossings by research aircraft in the northeast Atlantic in October 1993. For the age of plume crossings from 300 to 5600 s after the emission, the coefficients K_z and K_y are between 0 and $0.6 \text{ m}^2/\text{s}$ and between 5 and $20 \text{ m}^2/\text{s}$ correspondingly, and $K_z = 0.3 \text{ m}^2/\text{s}$ and $K_y = 12 \text{ m}^2/\text{s}$ may be adopted as the “best guess” values (Schumann *et al.*, 1995). The wind shear parameter s_z for the plume crossing periods and region are estimated to be between 0 and $(2-6)10^{-3} \text{ s}^{-1}$ from the meteorological data, such that s_z determines the plume horizontal dispersion, that is dominant after 1–2 h of plume “life”, but this is a less studied parameter of plume dispersion (Schumann and Konopka, 1994; Schumann *et al.*, 1995).

Table 1 presents the compositions of the initial exhaust from the engine and of the ambient air used in the model for the midday and midnight emission moments at 10 km level, 50°N latitude and in July. The gas concentrations in the ambient air are taken for the flight corridor (FC) with enhanced NO_x and H_2O and out side of it with the lower NO_x and H_2O of global clean air (CA) background. It should be noted, that while the (OH) background mixing ratio in Table 1 is twice as high in the FC than in the CA, the (HO_2) mixing ratio in FC is about twice as low than in CA evidently due to about 8 times more (NO_x) in FC ambient air. This regulates the (HO_2)/(OH) ratio mostly by the known reaction $\text{HO}_2 + \text{NO} \rightarrow \text{OH} + \text{NO}_2$. Variant E1 of emissions relates to B-747 type aircraft, while variant E2 corresponds to aircraft engines with smaller emission indices and fuel flows of B-737, B-767 type (Hoinka *et al.*, 1993; Schumann, 1994). Basic (B) variant is an intermediate between these two. The plume dispersion parameters due to this spread of their values, are taken at their extremes: maximal $K_z = 0.6 \text{ m}^2/\text{s}$; $K_y = 20 \text{ m}^2/\text{s}$; $s_z = 3 \cdot 10^{-3} \text{ s}^{-1}$ for the “wide” plume (WP), minimal $K_z = 0.1 \text{ m}^2/\text{s}$; $K_y = 5 \text{ m}^2/\text{s}$; $s_z = 0$ for the “narrow”

Table 1. Initial mixing ratios of aircraft emitted gases and their background values at the 10 km level for the calculated variants ($a(b) = a \cdot 10^b$)

Gas	Emissions (ppb)			Background (pptv)			
	Variants			Flight corridor		Clean air	
	B	E1	E2	Day	Night	Day	Night
NO	36.0	72.0	7.2	160.0	1.1 (–3)	23	5.1 (–4)
NO ₂	4.0	8.1	1.1	120.0	270.0	19	36
OH	1.4	0.0	1.2	0.4	9.2 (–4)	0.2	1.7 (–3)
HO ₂	0.0	0.0	0.0	2.9	8.5 (–2)	5.2	1.2 (–1)
CO	9.0	17.0	1.0	3.2 (4)	3.2 (4)	3.2 (4)	3.2 (4)
HNO ₃	0.0	0.0	0.0	3.2 (3)	3.2 (3)	490	490
H ₂ O	5.5 (3)	1.1 (4)	2.3 (3)	1.0 (8)	1.0 (8)	1.0 (7)	1.0 (7)

plume (NP), and the “best guess” $K_z = 0.3 \text{ m}^2/\text{s}$, $K_y = 12 \text{ m}^2/\text{s}$, $s_z = 0$, are all considered in parallel.

The time evolution of gas contents in the plume are related to their model diurnal course in the ambient air for the relevant latitude and month, as found in Karol and Ozolin (1994). For the flight corridor (FC) ambient air the enhanced NO_y and H_2O contents are used, connected with their estimates in Schumann (1994); Schumann and Konopka (1994) and Hoinka *et al.* (1993).

Only the gas phase photochemical reactions are considered here in the first study. In Karol and Ozolin (1994), it was shown that the heterogeneous reaction of N_2O_5 with H_2O on the ambient air aerosol surface and on one type of aerosols in the plume influence only the N_2O_5 evolution in the plume. The more detailed estimation of influence of heterogeneous reactions on aerosols in the plume will be considered later.

About 85 gas phase reactions for following 33 variable compounds of the most important gas “families” are taken into account and listed in Table 2: O_3 , O , $\text{O}(^1\text{D})$, H , OH , HO_2 , H_2O_2 , N , NO , NO_2 , NO_3 , N_2O_5 , HNO_2 , HNO_3 , HO_2NO_2 , Cl , ClO , HCl , HOCl , ClONO_2 , ClOO , Cl_2 , CH_3 , CH_3O_2 , $\text{CH}_3\text{O}_2\text{H}$, CH_3O , CH_2O , HCO , H_2 , N_2O , H_2O , CO , CH_4 . Numerical realization of the box model is the same as in Karol *et al.* (1994). In cross section vertical and horizontal plume diameters contain 15 and 30 grid points correspondingly at each time moment.

Special attention is paid to calculations of mass balance components at every time step. Absolute and relative inputs of photochemical sources-sinks, intensities of separate compounds and of their flow through the plume boundary into the rate of gas mass change in the plume cross section are evaluated for both plume models. In the box model the numerical estimation of the boundary flow by the formula (2) is checked by its variant with analytical time differentiation of the plume cross section area $S(t) = 4\pi\sigma_z(t)\sigma_y(t)$ with the expression (3) for $\sigma_z(t)$ and $\sigma_y(t)$.

For the Gaussian plume model the technique of plume cross boundary flow calculation by the usual diffusion formula leads to significant computational errors due to the rough numerical calculation of gas concentration gradient over the plume boundary, especially at the end of plume “life”, when this gradient is small. Therefore, the gas flow through the Gaussian plume boundary is evaluated as the balance residue of the gas content rate in the plume

Table 2. List of gas phase photochemical reactions included in model

1.	$\text{O}_2 + hv \rightarrow \text{O} + \text{O}$	14.	$\text{HNO}_2 + hv \rightarrow \text{OH} + \text{NO}$
2.	$\text{O}_3 + hv \rightarrow \text{O}(^1\text{D}) + \text{O}_2$	15.	$\text{CH}_2\text{O} + hv \rightarrow \text{H}_2 + \text{CO}$
3.	$\text{NO}_2 + hv \rightarrow \text{NO} + \text{O}$	16.	$\text{CH}_2\text{O} + hv \rightarrow \text{H} + \text{CHO}$
4.	$\text{NO}_3 + hv \rightarrow \text{NO}_2 + \text{O}$	17.	$\text{CH}_3\text{O}_2\text{H} + hv \rightarrow \text{CH}_3\text{O} + \text{OH}$
5.	$\text{NO}_3 + hv \rightarrow \text{NO} + \text{O}_2$	18.	$\text{NO} + hv \rightarrow \text{N} + \text{O}$
6.	$\text{N}_2\text{O}_5 + hv \rightarrow \text{NO}_2 + \text{NO}_3$	19.	$\text{OH} + \text{HNO}_3 \rightarrow \text{H}_2\text{O} + \text{NO}_3$
7.	$\text{HNO}_3 + hv \rightarrow \text{NO}_2 + \text{OH}$	20.	$\text{OH} + \text{CO} \rightarrow \text{CO}_2 + \text{H}$
8.	$\text{H}_2\text{O}_2 + hv \rightarrow \text{OH} + \text{OH}$	21.	$\text{O}_3 + \text{O} \rightarrow \text{O}_2 + \text{O}_2$
9.	$\text{HCl} + hv \rightarrow \text{H} + \text{Cl}$	22.	$\text{O}(^1\text{D}) + \text{N}_2\text{O} \rightarrow \text{N}_2 + \text{O}_2$
10.	$\text{ClONO}_2 + hv \rightarrow \text{Cl} + \text{NO}_3$	23.	$\text{O}(^1\text{D}) + \text{N}_2\text{O} \rightarrow \text{NO} + \text{NO}$
11.	$\text{HOCl} + hv \rightarrow \text{Cl} + \text{OH}$	24.	$\text{O}(^1\text{D}) + \text{H}_2\text{O} \rightarrow \text{OH} + \text{OH}$
12.	$\text{HO}_2\text{NO}_2 + hv \rightarrow \text{HO}_2 + \text{NO}_2$	25.	$\text{O}(^1\text{D}) + \text{CH}_4 \rightarrow \text{CH}_3 + \text{OH}$
13.	$\text{Cl}_2 + hv \rightarrow \text{Cl} + \text{Cl}$	26.	$\text{O}(^1\text{D}) + \text{H}_2 \rightarrow \text{OH} + \text{H}$
		27.	$\text{O}(^1\text{D}) + \text{N}_2 \rightarrow \text{O} + \text{N}_2$
		28.	$\text{O}(^1\text{D}) + \text{O}_2 \rightarrow \text{O} + \text{O}_2$
		29.	$\text{O} + \text{OH} \rightarrow \text{O}_2 + \text{H}$
		30.	$\text{O} + \text{HO}_2 \rightarrow \text{OH} + \text{O}_2$
		31.	$\text{OH} + \text{HO}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$
		32.	$\text{OH} + \text{O}_3 \rightarrow \text{HO}_2 + \text{O}_2$
		33.	$\text{OH} + \text{OH} \rightarrow \text{H}_2\text{O} + \text{O}$
		34.	$\text{OH} + \text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{HO}_2$
		35.	$\text{OH} + \text{H}_2 \rightarrow \text{H}_2\text{O} + \text{H}$
		36.	$\text{HO}_2 + \text{HO}_2 \rightarrow \text{H}_2\text{O}_2 + \text{O}_2$
		37.	$\text{HO}_2 + \text{O}_3 \rightarrow \text{OH} + 2\text{O}_2$
		38.	$\text{N} + \text{O}_2 \rightarrow \text{NO} + \text{O}$
		39.	$\text{N} + \text{NO} \rightarrow \text{N}_2 + \text{O}$
		40.	$\text{O} + \text{NO}_2 \rightarrow \text{NO} + \text{O}_2$
		41.	$\text{O} + \text{NO}_3 \rightarrow \text{O}_2 + \text{NO}_2$
		42.	$\text{O}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{O}_2$
		43.	$\text{NO} + \text{HO}_2 \rightarrow \text{NO}_2 + \text{OH}$
		44.	$\text{NO} + \text{NO}_3 \rightarrow \text{NO}_2 + \text{NO}_2$
		45.	$\text{O}_3 + \text{NO}_2 \rightarrow \text{NO}_3 + \text{O}_2$
		46.	$\text{OH} + \text{CH}_4 \rightarrow \text{CH}_3 + \text{H}_2\text{O}$
		47.	$\text{OH} + \text{CH}_2\text{O} \rightarrow \text{H}_2\text{O} + \text{CHO}$
		48.	$\text{CH}_3\text{O} + \text{O}_2 \rightarrow \text{CH}_2\text{O} + \text{HO}_2$
		49.	$\text{CHO} + \text{O}_2 \rightarrow \text{CO} + \text{HO}_2$
		50.	$\text{CH}_3\text{O}_2 + \text{CH}_3\text{O}_2 \rightarrow \text{CH}_3\text{O} + \text{CH}_3\text{O}$
		51.	$\text{CH}_3\text{O}_2 + \text{NO} \rightarrow \text{CH}_3\text{O} + \text{NO}_2$
		52.	$\text{CH}_3\text{O}_2 + \text{HO}_2 \rightarrow \text{CH}_3\text{O}_2\text{H} + \text{O}_2$
		53.	$\text{Cl} + \text{O}_3 \rightarrow \text{ClO} + \text{O}_2$
		54.	$\text{Cl} + \text{H}_2 \rightarrow \text{HCl} + \text{H}$
		55.	$\text{Cl} + \text{CH}_4 \rightarrow \text{HCl} + \text{CH}_3$
		56.	$\text{Cl} + \text{CH}_2\text{O} \rightarrow \text{HCl} + \text{CHO}$
		57.	$\text{Cl} + \text{H}_2\text{O}_2 \rightarrow \text{HCl} + \text{HO}_2$
		58.	$\text{Cl} + \text{HOCl} \rightarrow \text{Cl}_2 + \text{OH}$
		59.	$\text{Cl} + \text{HO}_2 \rightarrow \text{HCl} + \text{O}_2$
		60.	$\text{Cl} + \text{HO}_2 \rightarrow \text{OH} + \text{ClO}$
		61.	$\text{Cl} + \text{NO}_3 \rightarrow \text{ClO} + \text{NO}_2$
		62.	$\text{Cl} + \text{ClOO} \rightarrow \text{Cl}_2 + \text{O}_2$
		63.	$\text{Cl} + \text{ClOO} \rightarrow \text{ClO} + \text{ClO}$
		64.	$\text{ClO} + \text{O} \rightarrow \text{Cl} + \text{O}_2$
		65.	$\text{ClO} + \text{NO} \rightarrow \text{Cl} + \text{NO}_2$
		66.	$\text{ClO} + \text{HO}_2 \rightarrow \text{HOCl} + \text{O}_2$
		67.	$\text{OH} + \text{HCl} \rightarrow \text{H}_2\text{O} + \text{Cl}$
		68.	$\text{OH} + \text{HOCl} \rightarrow \text{H}_2\text{O} + \text{ClO}$
		69.	$\text{HNO}_2 + \text{OH} \rightarrow \text{HO}_2 + \text{NO}_2$
		70.	$\text{HO}_2\text{NO}_2 \rightarrow \text{HO}_2 + \text{NO}_2$
		71.	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$
		72.	$\text{ClOO} \rightarrow \text{Cl} + \text{O}_2$
		73.	$\text{O} + \text{O}_2 (+\text{M}) \rightarrow \text{O}_3$
		74.	$\text{H} + \text{O}_2 (+\text{M}) \rightarrow \text{HO}_2$
		75.	$\text{OH} + \text{OH} (+\text{M}) \rightarrow \text{H}_2\text{O}_2$
		76.	$\text{O} + \text{NO} (+\text{M}) \rightarrow \text{NO}_2$
		77.	$\text{O} + \text{NO}_2 (+\text{M}) \rightarrow \text{NO}_3$
		78.	$\text{OH} + \text{NO}_2 (+\text{M}) \rightarrow \text{HNO}_3$
		79.	$\text{HO}_2 + \text{NO}_2 (+\text{M}) \rightarrow \text{HO}_2\text{NO}_2$
		80.	$\text{NO}_2 + \text{NO}_3 (+\text{M}) \rightarrow \text{N}_2\text{O}_5$
		81.	$\text{Cl} + \text{O}_2 (+\text{M}) \rightarrow \text{ClOO}$
		82.	$\text{ClO} + \text{NO}_2 (+\text{M}) \rightarrow \text{ClONO}_2$
		83.	$\text{CH}_3 + \text{O}_2 (+\text{M}) \rightarrow \text{CH}_3\text{O}_2$
		84.	$\text{NO} + \text{OH} (+\text{M}) \rightarrow \text{HNO}_2$

cross section. Special studies reveal that this evaluation has an error of about 10–30% and is more precise than the usual diffusion formula.

3 Results of calculations

The calculated time evolution of ozone, $\text{NO}_x = \text{NO} + \text{NO}_2$, HO_x , HNO_2 , HNO_3 , HO_2NO_2 component mixing ratios in the box model of the plume relative to their background diurnal courses are very close to these presented in Karol and Ozolin (1994), for 10 km level and they are not repeated here. As in the latter paper the ozone depletion, up to a maximum of 20%, occurs during the first minutes of plume life in the jet and vortex regimes with little air exchange through the plume boundary. This ozone change does not influence, in practice, the plume composition evolution in the plume dispersion regime (Karol *et al.*, 1994; Veenstra and Beck, 1994).

Figure 1 compares the NO , OH and HNO_2 mixing ratio distributions along the plume axis horizontal cross radius in the box and in the 2D Gaussian plume models for the basic variant in the flight corridor (FC) at 25 min after the midday injection. Figure 1 shows that NO and HNO_2 are spreading from the plume, but the OH deficit in the plume must be replenished from the outside HO_x . The spatially averaged compound content over the plume horizontal radius of 400 m in the 2D plume model is evidently close to its value in the box model. The HNO_3 and HO_2NO_2 distributions over the centered horizontal Gaussian plume cross section are close to their concentrations in the box model, indicating the much smaller effects of their transport through the plume boundary on evolution of their content in the plume. The same relations hold during the whole plume “life” for most of gases as may be

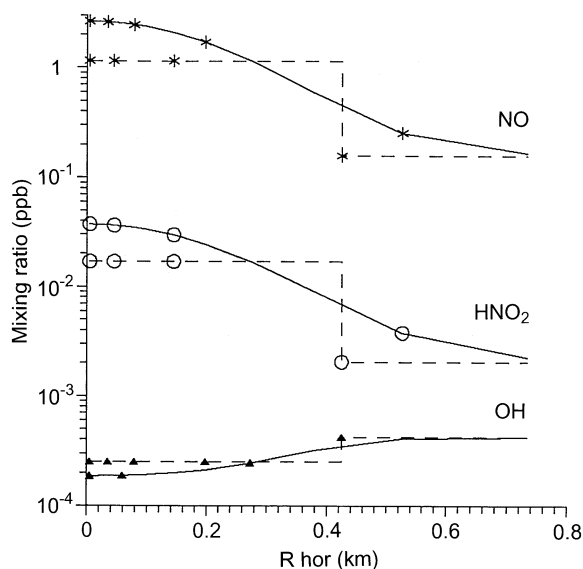


Fig. 1. NO , HNO_2 and OH mixing ratio distributions along the plume axis horizontal cross radius in the box (*dashed lines*) and the 2-D Gaussian (*solid lines*) plume models for the basic variant in the flight corridor (FC) at 25 min after the midday injection

seen from Fig. 2, which presents the time evolution of HNO_2 mixing ratios: at the Gaussian plume axis, at plume edge and integrated over the plume cross section area related to HNO_2 mixing ratio in the box model for the same basic variant. These relations become close to unit (gas mixing ratios in the Gaussian plume tend to uniformity in plume cross section) at the end of plume “life” considered here. The rate of this tendency is different for various gases: for instance it is less rapid for NO_x and HO_2 .

Figure 3 demonstrates plume model sensitivity to atmospheric dispersion parameters and compares the calculated (NO) mixing ratio time evolution in the box plume

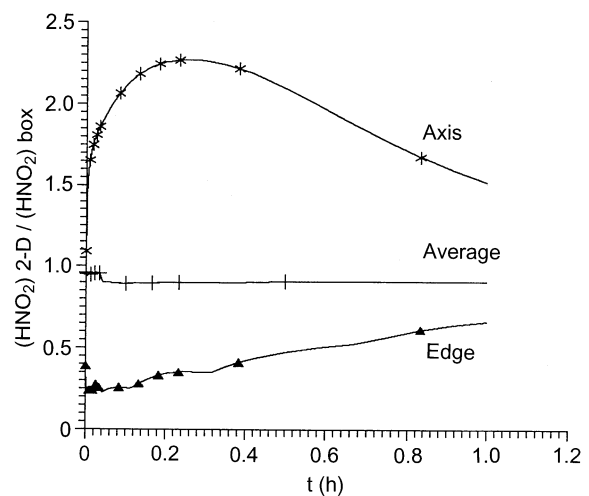


Fig. 2. Time evolution of the HNO_2 mixing ratio: at the Gaussian plume axis; at this plume edge; and integrated over the plume cross section area (*average*)—relative to the HNO_2 mixing ratio in the box model for the same basic variant

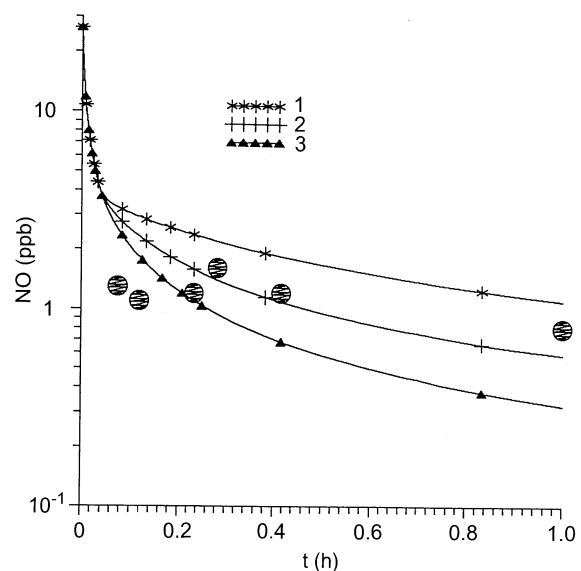


Fig. 3. Time evolution of the (NO) mixing ratio in the model box: 1, narrow plume; 2, basic; 3, wide plume, and circles, measured (NO) in cross section of various exhaust plumes over the eastern Atlantic (Schlager *et al.*, 1994)

with measured (NO) in cross sections of various exhaust plumes by the research aircraft over the eastern Atlantic (Schlager *et al.*, 1994). It is evident, that (NO) in the narrow plume box model is very different from the real data for plume age beyond 10 min, but then basic box model results are close to measurements. It may be concluded from Fig. 3, that during the plume evolution (NO) mixing ratio in the box plume becomes less dependent from σ_{oz} , σ_{oy} and more sensitive to atmosphere dispersion parameters K_y , K_z .

Figures 4, 5 and 6 compare some variants of the time evolution of NO, OH, HNO₃ balance components in the

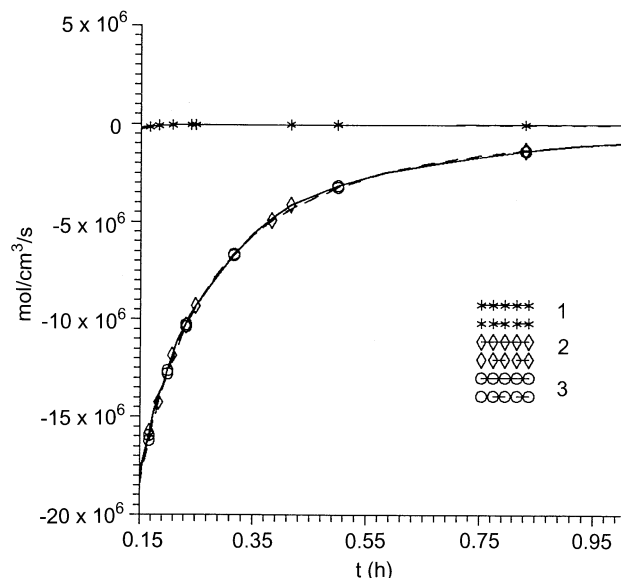


Fig. 4. Time evolution of NO_x content rate of change in cross section of box (*dashed line*) and Gaussian (*solid line*) model plumes; 1, due to photochemical processes; 2, due to plume trans-boundary air transport; 3, total

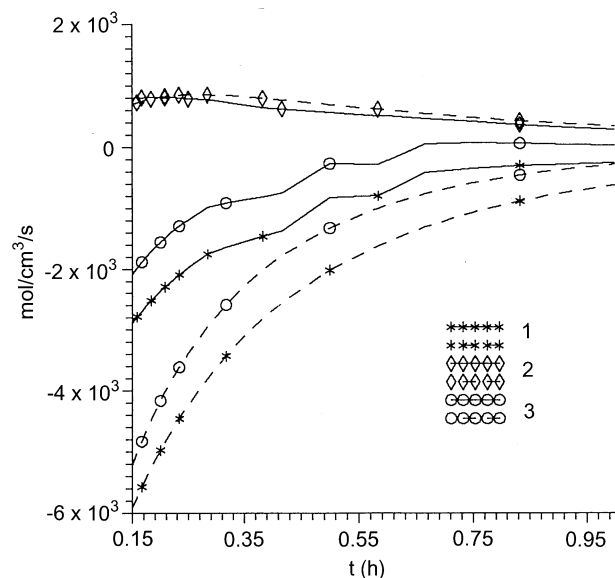


Fig. 5. Same as in Fig. 4 but for OH evolution

cross sections of box and Gaussian plume models. In both the models the inputs of photochemical transformations and of the air exchange through plume boundary into the rate of gas cross section content change have opposite signs for most of the gases and their maxima are in the first minutes of the plume dispersion regime. These maxima and input levels are almost the same in the box and in the Gaussian model for all gases at all times. Only the photochemical consumption of the short-lived HO_x components and, therefore, their total change rates in the plume cross section are higher in the box than in the Gaussian model (see Fig. 5). The total gas content change rates in the plume cross section are much less and almost coincide in both models for all gases and for variants with the same parameters. Photochemical transformations and plume boundary air transport components of the change rates always act in opposite directions and partly compensate each other. Therefore, the differences between the plume box and Gaussian models in the photochemical transformation and in the gas transport rates through the plume boundary, being not very significant separately, produce almost the same total gas content change in the plume cross sections during the whole plume “life” period. This important conclusion is valid for all the gases considered and confirms the use of the simpler and handier box model for studying the general effects of the aircraft exhaust plume, of its dispersion and photochemistry on the ambient atmosphere. Only the plume box models are used in most of the studies of this effect (Danilin *et al.*, 1992; Karol *et al.*, 1994; Plejěl *et al.*, 1994; Veenstra and Beck, 1994 and others), and the above results give them an additional basic support.

Only the results from the box model simulations will be presented below. Figures 7–10 demonstrate the time evolution of NO, NO_x, HNO₂, HNO₃, HO₂NO₂ contents over their backgrounds relative to total odd nitrogen NO_y = NO_x + NO₃ + HNO₂ + HNO₃ + HO₂NO₂ + 2N₂O₅ amount in the plume cross sections and their

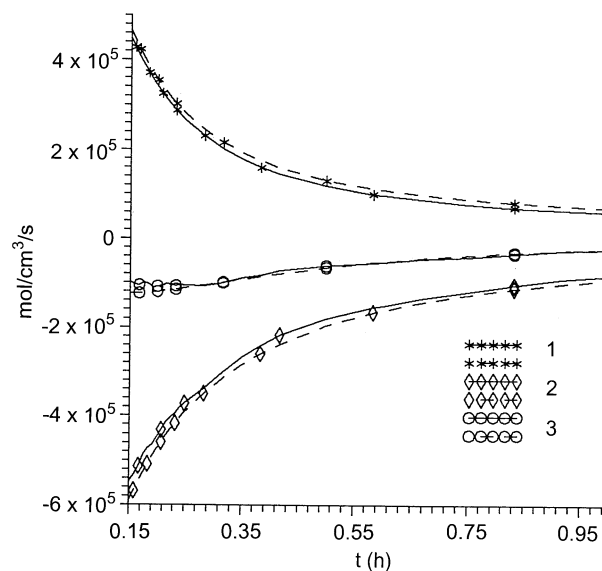


Fig. 6. Same as in Fig. 4 but for HNO₃ evolution

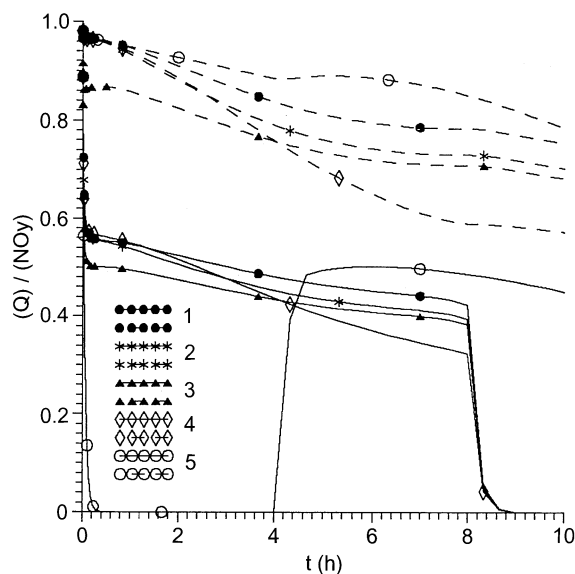


Fig. 7. Time evolution of NO (solid lines) and NO_x (dashed lines) relative to NO_y contents in the plume box model over their backgrounds for B midday emission in FC and for variants with: 1, basic plume; 2, wide plume; 3, basic plume with five times increased emission of OH; 4, basic plume in clean air; 5, midnight basic injection in the flight corridor

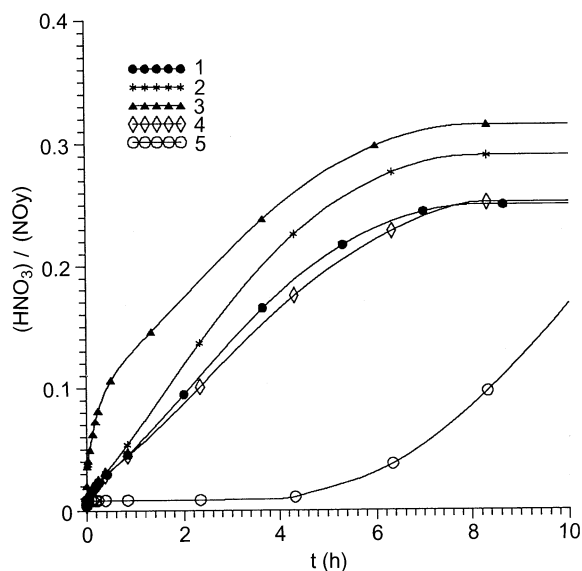


Fig. 8. Same as in Fig. 7 but for HNO₃ evolution

sensitivities to some of model parameter variations. The fivefold increase of OH in exhausts against the basic variant of emissions results in about 10% reduction of NO_x/NO_y, 7% reduction of NO/NO_y during the entire 10 h of model plume calculations and in 2–5% increase of HNO₂/NO_y ratios during the first 2 h of plume dispersion period. About 7% of HNO₃/NO_y and 1% of HO₂NO₂/NO_y increase persist during the entire 10 h of plume “life”. In the case of nighttime emission there is minimal NO_x oxidation and HNO₃ formation during the same period.

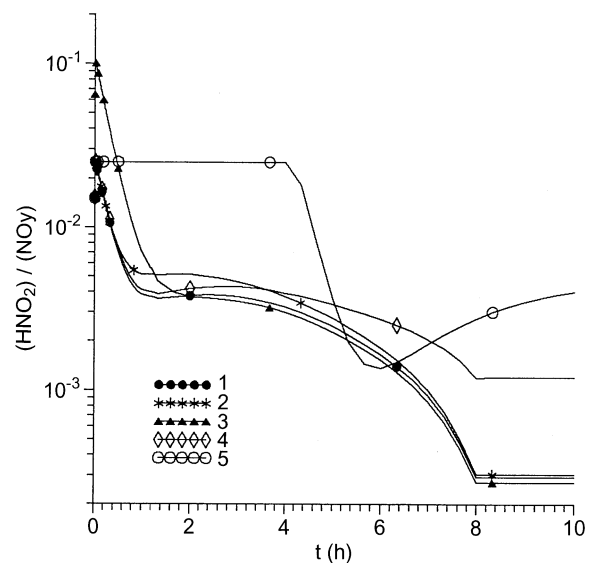


Fig. 9. Same as in Fig. 7 but for HNO₂ evolution

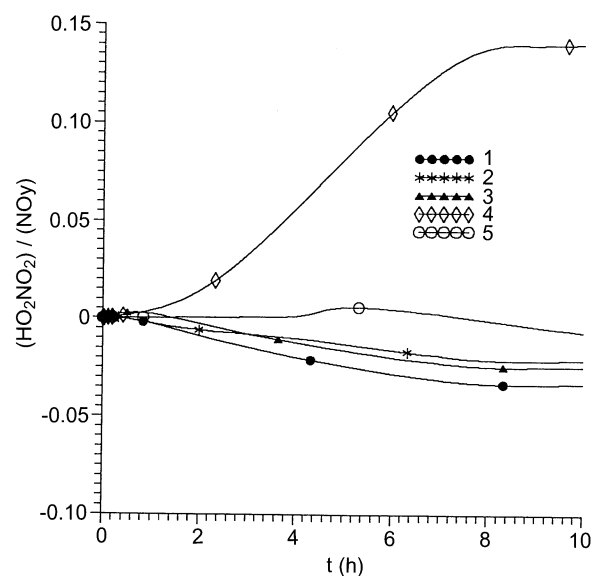


Fig. 10. Same as in Fig. 7 but for HO₂NO₂ evolution

Some increases of the ambient air transport into the wide plume in comparison with the basic variant result in 5% reduction of NO_x/NO_y ratio and in 4% increase of HNO₃/NO_y at the end of the 10 h plume period. The maximal NO_x/NO_y reduction to almost 50% of initially emitted occurs by the end of the same period in the plume in the clean ambient air. This is evidently due to external air inflow into the plume from the clean air background which is low in NO_x and due to enhanced outflow of emitted NO_x from the plume into the low in NO_x external air (see Table 1). The NO_x concentration difference inside and outside of the plume is lower, and HO_x external content is higher in the flight corridor in comparison to clean air conditions, which leads to the differences noted in the plume NO_x oxidation rate.

Figures 8 and 9 reveal the significant HNO_3/NO_y and HNO_2/NO_y enhancements during the plume “life” for variant with increased OH emission, when emitted OH oxidates NO into HNO_2 and NO_2 into HNO_3 during the first few minutes. Then HNO_2 formed disappears in one hour, but HNO_3 remains to the end of the plume “life”, attaining a maximum of about one third of NO_y content. In the first hours of the plume dispersion during the day OH and HO_2 enter into the plume with the entrained air, and they oxidate NO_x in the plume forming the HNO_2 , HNO_3 , HO_2NO_2 content excess there. During the night the opposite happens: the emitted HO_x radicals leave the plume and with low external HO_x produce the low ratio HNO_3/NO_y and, together with the low ratio HNO_2/NO_y , conserve these till sunrise. Moderate enhancement of HNO_3/NO_y occurs in the wide plume after about 0.5 h from emission, and is due to air exchange increase through the plume boundary, bringing more external OH into the plume for NO_x oxidation. This enhanced air exchange influences the HNO_3 formation rate more than the plume and the external OH level difference in the flight corridor and in the clean air conditions (Table 1), being opposite to the distinct influence of these factors on the NO_x/NO_y time evolution in Fig. 7.

The $(\text{HO}_2\text{NO}_2)/(\text{NO}_y)$ time evolution shown in Fig. 10 is significantly different from the HNO_3/NO_y ratio behavior and opposite in the flight corridor (FC) and in the clean air (CA) environment. An intensive formation of HO_2NO_2 in the plume in one hour after the emission in the latter case is due to inflow of external air with relatively high HO_2 concentration (see Table 1). In the wide plume (WP), with enhanced air exchange through its boundary, the $\text{HO}_2\text{NO}_2/\text{NO}_y$ increase is more pronounced. But in the FC, with lower HO_2 and relatively high OH ambient concentrations, $\text{HO}_2\text{NO}_2/\text{NO}_y$ inside the plume may be slightly deficient relative to its background value during the day, as HO_2NO_2 photochemical sink prevails over its source intensities. These HO_2NO_2 evolutions do not influence the final NO_y component distributions noticeably in the plume.

4 Discussion and conclusions

The results presented of aircraft exhaust plume evolution during the first 10 h of its “life” by box and Gaussian plume approximations are shown to be in line with known differences in the plume gas concentration distributions, described by these widely used plume models. They also show close proximity in presenting the integral gas content time evolution in the cross section of plume with the same parameters and external conditions. The resulting integral gas content rates of change in the plume cross section are close to each other in both plume models. This important conclusion confirms the wide use of the box plume model in studies of photochemical transformations of emitted atmospheric pollutants.

These simulations support our previous work (Karol *et al.*, 1994) and our conclusions on the substantial influence of ambient air composition, of this air entrainment rate into the plume on the plume gas composition evolution,

and on the NO_x oxidation rate into HNO_3 , HO_2NO_2 in the plume in particular. As was indicated in Karol and Ozolin (1994) the rates of NO_x oxidation and HNO_3 formation in the plume are roughly proportional to the OH (approximately to H_2O and O_3) mixing ratio in the clean ambient air with low background NO_x (see Fig. 11). The enhanced H_2O , expected over the North Atlantic FC, favors an increased NO_x oxidation rate in the plume. At the same time the enhanced NO_x background concentrations in the flight corridor reduce the NO_x in NO_y share depletion to the end of the plume “life” as indicated in Fig. 7. The resulting partial compensation of these two processes would obviously depend on the NO_x and H_2O concentration levels in the FC air, which may obviously vary significantly due to superposition of aircraft exhaust plumes traces there (Schuman and Konopka, 1994). The quantitative estimation of NO_x reduction in the plume to the end of its “life” in the FC would become more complicated with frequent cases of exhaust plume intersections or of plumes imbedding into the expanded plume remnants of previous aircraft cruising along the same trajectory. In the ambient clean air environment the enhanced

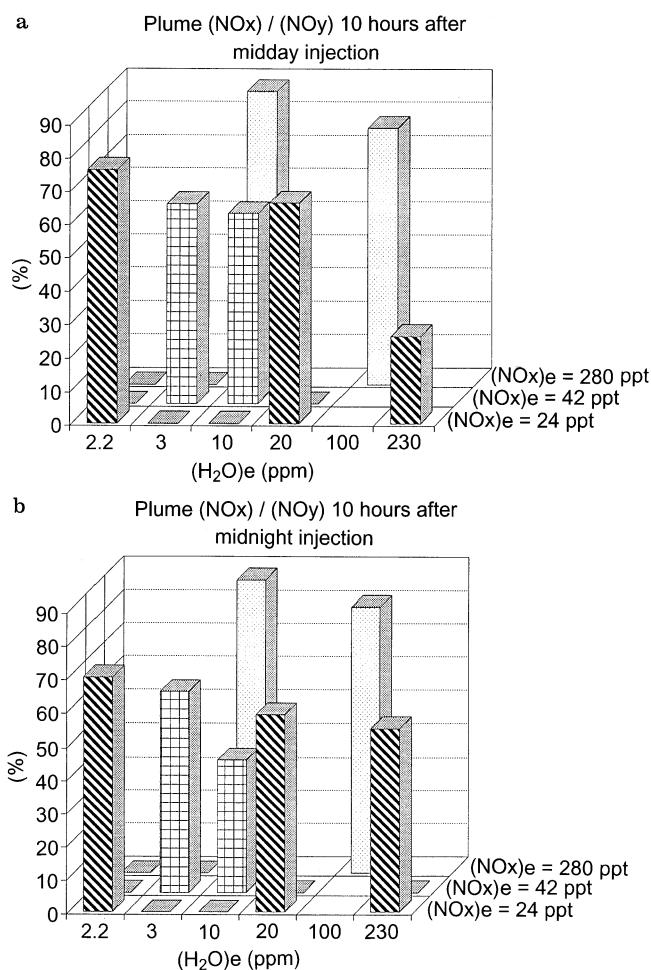


Fig. 11a, b. The box plume $(\text{NO}_x)/(\text{NO}_y)$ ratio (%) dependence at 10 h after **a** the midday and **b** midnight injection from the $(\text{H}_2\text{O})_e$ and $(\text{NO}_x)_e$ mixing ratios in the external air

$\Delta(\text{NO}_x)/\Delta(\text{NO}_y)$ drop in the 10 h period (Fig. 7) is due to the entrainment into the plume of the low in NO_x external air. Because this external air has low OH, but high HO_2 (Table 1), it does not influence the HNO_3 , but enhances the HO_2NO_2 formation rate in the plume as may be seen in Figs. 8 and 10. Figure 8 shows, that the enhanced air transport through the “wide” plume boundary in the FC does not influence the (HNO_3) share in the plume (NO_y) due to modest external OH entering into the plume for NO_x oxidation.

Figures 11 and 12 demonstrate the connection of the NO_x and HNO_3 shares in NO_y to the end of plume “life” with (NO_x) and (H_2O) mixing ratio in the ambient air, which partly determines the OH concentration there. For the ambient (H_2O) = const., the NO_x/NO_y ratio share increases for background (NO_x) rise, and for ambient (NO_x) = const. the same ratio declines for external (H_2O) and (OH) enhancement, reflecting the already found compensating influence of the background air composition on the NO_x oxidation rate in the plume. A connection for the HNO_3/NO_y ratio with external (NO_x)_e and (H_2O)_e over the 10 h plume life is presented at Fig. 12. The almost linear increase of this ratio for ambient (H_2O) and (OH) rise is understandable. The sharp decline of the HNO_3

share in NO_y in the plume for flight corridor NO_x levels in the ambient air and for clean air (H_2O) and HO_x concentrations is the evident result of the very low NO_x oxidation rate in the plume in these conditions (only about 10% of emitted NO_x is oxidized, see Fig. 11).

All these results suggest the feasibility of introducing into the assessment of aircraft emission atmospheric effects the concept of the Effective Emission Index (EEI) as a generalization and development of the well-known Emission Index (EI), which indicates the NO_x emitted from engine in terms of grams per kilogram of fuel burned (Schuman, 1994; Schulte and Schlanger, 1994; WMO, 1995). Only EEI and not EI has to be used in evolution of NO_x atmospheric sources from the aircraft exhausts in global modeling of their atmospheric effects.

In addition to the model calculations and laboratory testing, the actual EI is determined now also by NO_x measurements in the aircraft plume crossings by research aircraft (Schulte and Schlager, 1994; Fahey *et al.*, 1995). The EEI evaluation and its sensitivity studies may be carried out not only by extensive modeling, but also by actual aircraft measurements of NO_y component contents in exhaust plumes in various environments and flight conditions. EEI evaluation and research is a large complicated task and this study is only the first step in this direction.

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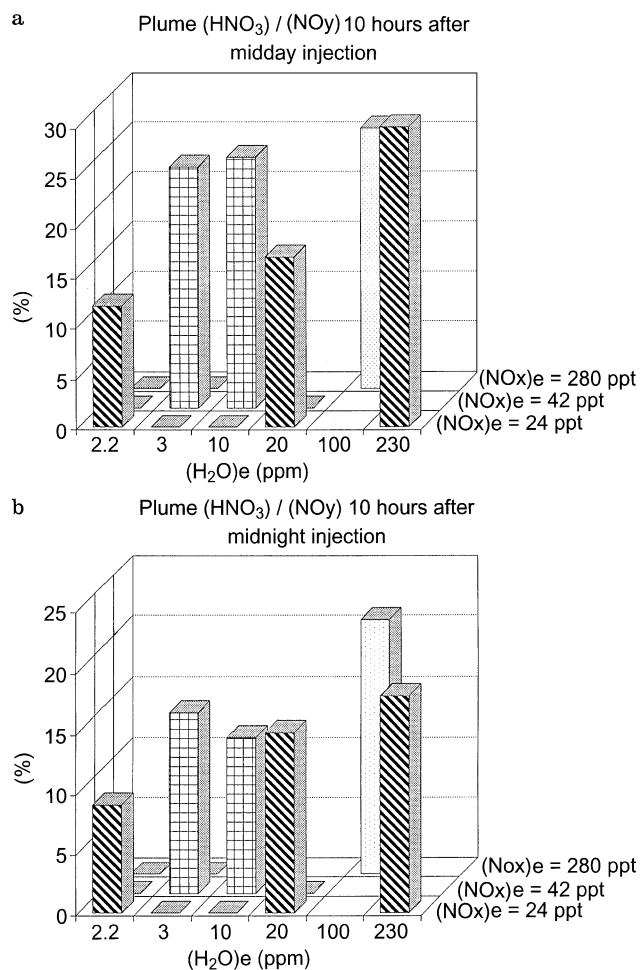


Fig. 12a, b. Same as in Fig. 11 but for (HNO_3)/(NO_y)

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