



Quantification of the depletion of ozone in the plume of Mount Etna

L. Surl¹, D. Donohue^{1,*}, A. Aiuppa^{2,3}, N. Bobrowski⁴, and R. von Glasow¹

¹Centre for Ocean and Atmospheric Sciences, School of Environmental Sciences, University of East Anglia, Norwich, NR4 7TJ, UK

²DiSTeM, Università di Palermo, 90123 Palermo, Italy

³Istituto Nazionale di Geofisica e Vulcanologia, Sezione di Palermo, Via La Malfa 153, 90146 Palermo, Italy

⁴Institut für Umweltphysik, Universität Heidelberg, Im Neuenheimer Feld 229, 69120 Heidelberg, Germany

* now at: Department of Chemistry, Lawrence University, Appleton, Wisconsin, 54911, USA

Correspondence to: L. Surl (l.surl@uea.ac.uk)

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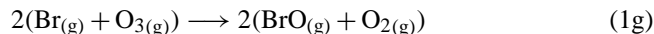
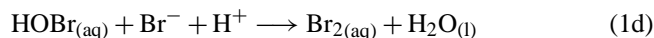
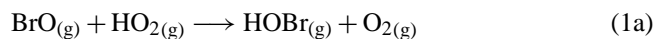
Abstract. Volcanoes are an important source of inorganic halogen species into the atmosphere. Chemical processing of these species generates oxidised, highly reactive, halogen species which catalyse considerable O₃ destruction within volcanic plumes. A campaign of ground-based in situ O₃, SO₂ and meteorology measurements was undertaken at the summit of Mount Etna volcano in July/August 2012. At the same time, spectroscopic measurements were made of BrO and SO₂ columns in the plume downwind.

Depletions of ozone were seen at all in-plume measurement locations, with average O₃ depletions ranging from 11–35 nmol mol⁻¹ (15–45 %). Atmospheric processing times of the plume were estimated to be between 1 and 4 min. A 1-D numerical model of early plume evolution was also used. It was found that in the early plume O₃ was destroyed at an approximately constant rate relative to an inert plume tracer. This is ascribed to reactive halogen chemistry, and the data suggests the majority of the reactive halogen that destroys O₃ in the early plume is generated within the crater, including a substantial proportion generated in a high-temperature “effective source region” immediately after emission. The model could approximately reproduce the main measured features of the ozone chemistry. Model results show a strong dependence of the near-vent bromine chemistry on the presence or absence of volcanic NO_x emissions and suggest that near-vent ozone measurements can be used as a qualitative indicator of NO_x emission.

1 Introduction

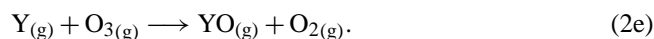
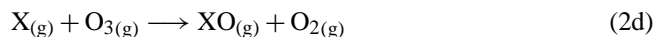
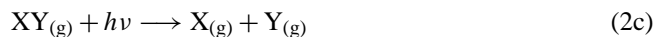
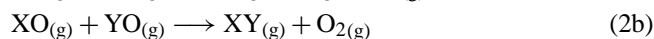
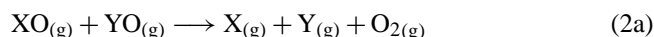
Volcanoes are known to be a major source of climatically important species, trace gases, and aerosol into the atmosphere, both through explosive eruptions and sustained quiescent degassing (e.g. von Glasow et al., 2009). The largest gaseous components of most eruptions are water, carbon dioxide and sulphur compounds, followed by hydrogen halides (Symonds et al., 1994). These volcanogenic halides are known to present environmental hazards upon deposition to soils (e.g. Delmelle et al., 2003). Emissions of bromide, while weaker and less well studied than those of fluoride and chloride (Pyle and Mather, 2009), are of interest to atmospheric chemistry due to the conversion of HBr to very reactive ozone-depleting bromine species (Bobrowski et al., 2003, 2007; Bobrowski and Platt, 2007; Gerlach, 2004; von Glasow et al., 2009; von Glasow, 2010; Oppenheimer et al., 2006; Kern et al., 2009; Roberts et al., 2009, 2014; Kelly et al., 2013).

The *bromine explosion* is an autocatalytic chemical reaction cycle which is known to convert bromine from HBr to very reactive forms (Wennberg, 1999). It is considered to be the reason for detections of elevated BrO associated with depleted O₃ (e.g. von Glasow et al., 2009; Saiz-Lopez and von Glasow, 2012).



There is no equivalent *chlorine explosion* as HCl is less reactive than HBr (see von Glasow et al., 2009).

In the troposphere, reactive halogen species catalyse ozone destructive cycles (e.g. Saiz-Lopez and von Glasow, 2012) (X and Y are halogen atoms):



Reactive halogen chemistry and related O₃ depletion has been studied in detail in the polar and marine boundary layers and over salt lakes (see Saiz-Lopez and von Glasow, 2012, for an overview).

BrO is detectable through spectroscopy, and detection of elevated levels are used as an indicator for the occurrence of reactive bromine chemistry. BrO was first detected in a volcanic plume by Bobrowski et al. (2003), and has, to date, been detected in the plumes of about 20 different volcanoes. A good overview of ground-based measurements of BrO in volcanic plumes can be found in Boichu et al. (2011) and references therein. A systematic study of satellite data by Hörmann et al. (2012) greatly extended the data set of known BrO events detected via satellite measurements.

In both satellite and ground-based studies, in-plume ratios of BrO to SO₂ have been found to be of the order of 10⁻⁵–several 10⁻⁴ once the plume has been transported a few km from the source (e.g. Bobrowski et al., 2007; Boichu et al., 2011; Hörmann et al., 2012). Measurements and modelling studies of in-plume BrO/SO₂ ratios have shown BrO to be either present at low levels or undetectable within 1 km of volcanic vents whilst the BrO/SO₂ ratio a few km downwind of the same vent is measured as up to several 10⁻⁴ (e.g. Bobrowski et al., 2007; Kern et al., 2009). This suggests the formation of BrO is occurring within the time taken for this downwind transport. SO₂ is widely used as a tracer for volcanic emissions due to its high concentration in volcanic plumes, low reactivity, and easily detectable spectroscopic signal (McGonigle et al., 2004).

Volcanoes are known to emit other halogens (F, Cl, I). The HF molecule is expected to be too stable for any in-plume

Table 1. Table of reported depletions of O₃ in tropospheric volcanic plumes.

Volcano	Measurement platform	Reference
Augustine	airborne	Vance et al. (2010)
Erebus	airborne	Oppenheimer et al. (2010) Boichu et al. (2011)
Etna	ground-based	Vance et al. (2010)
Eyjaflajallajökull	airborne	Vance et al. (2010) Schumann et al. (2011)
St. Helens	airborne	Hobbs et al. (1982)
Redoubt	airborne	Kelly et al. (2013)
Sakurajima	ground-based	Lee et al. (2005)

reactive F chemistry to occur (von Glasow et al., 2009), whilst the magnitudes of volcanic HI emissions are probably too low to have a significant effect on plume chemistry (Aiuppa et al., 2005a). Measurements of oxidised Cl in volcanic plumes are varied, having been observed at Sakurajima (Lee et al., 2005), Etna (Bobrowski et al., 2007; General et al., 2014), Puyehue-Cordón Caulle (Theys et al., 2014), and Soufrière Hills (Donovan et al., 2014). However a DOAS investigation by Kern et al. (2009) specifically looking for oxidised Cl did not detect any in the plume of Masaya Volcano. It is unknown whether this reflects natural variability or measurement issues. See Roberts et al. (2009) for a discussion of reactive chlorine formation in volcanic plumes.

The conceptual model of halogen-O₃ chemistry within volcanic plumes is complicated by the very high temperatures of the plume immediately after emission and mixing with the atmosphere. Such high temperature mixtures are believed to reach or approach thermodynamic equilibrium (Symonds et al., 1994), resulting in the production of non-negligible quantities of oxidised halogen species which may begin the autocatalytic cycles described above (Gerlach, 2004; Bobrowski et al., 2007; Martin et al., 2009). Because of the hazards involved, this “effective source region” within the crater (Bobrowski et al., 2007) cannot be investigated directly. Instead, inventories of volcanic emission species are based mostly on samples of fumarole emissions or crater-rim measurements (Gerlach, 2004). Processes occurring in the effective source region are inferred from these measurements, theory, and measurements of the plume further downwind.

While BrO (and, with greater difficulty, OClO) can be measured remotely using passive spectroscopic techniques, direct measurements of O₃ depletion within volcanic plumes have to be made in situ due to the large atmospheric O₃ column, and as a result the existing body of literature on the depletion of ozone within volcanic plumes is small, see Table 1. Active DOAS measurement of O₃ at a volcano is possible, however it presents significant difficulties in distinguishing the plume from background air. To the authors’ knowledge

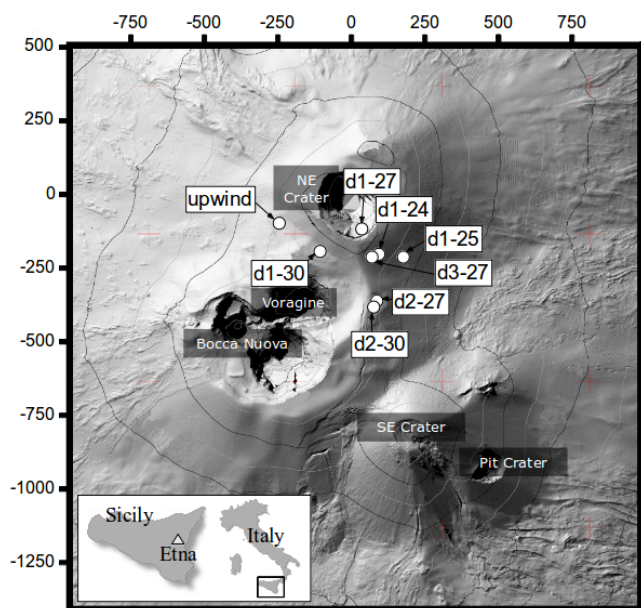


Figure 1. Map of the sampling sites and summit craters. Terrain from Neri et al. (2009). Coordinates are metres displacement from the NE crater.

the only published active DOAS O_3 measurement at a volcano is Kern et al. (2009). In this investigation, no O_3 depletion could be determined – the authors of this publication attribute this to the difficulties of isolating a plume O_3 signal from that of the background air.

This study expands this small collection of O_3 observations. In addition, the O_3 depletion data collected are associated with simultaneous measurements of an inert plume tracer (SO_2). To the authors' knowledge, no prior data set exists of simultaneous measurements of O_3 depletion and a tracer species within 1 km of volcanic vents. Wind speeds have also been measured, such that processing time can be determined for every datapoint. As in-plume halogen and O_3 chemistry are closely linked, from these data we can investigate both qualitatively and quantitatively the fast chemical processes occurring within the early plume.

2 Campaign and data processing

2.1 Location

The measurement campaign was conducted at Mount Etna, Sicily, Italy (37.7° N, 15.0° E) with the majority of measurements being taken between 24 July and 2 August 2012. Mount Etna is an active stratovolcano with a peak elevation of about 3300 m. At the time of measurement, the volcano was in a passively degassing phase with three active vents (North-East, Voragine and Bocca Nuova) within our survey area. Small Strombolian eruptions and an inner-crateric lava flow were observed at Bocca Nuova but explosive ejecta only

rarely surpassed the crater rim. Due to the vicinity of the Voragine and Bocca Nuova craters it was not possible to distinguish between their plumes. Therefore in this analysis we have considered them to be a single source, which we define as the “central craters”.

Most of the measurements were taken near the crater rim at elevations significantly above the planetary boundary layer. Due to logistical constraints we took these measurements between 08:40 and 12:00 UTC (10:40–14:00 local time). The sky was cloud-free on all days of measurement. The ground-based remote sensing measurements of the plume were taken from locations further downwind with plume ages of up to 30 min.

On all days of near-crater measurements we took measurements at a fixed site that was upwind of all summit craters and at 1–3 sites in the plume at varying distances from the craters (Fig. 1). This was done to sample plumes at different processing times. Sampling times at the sites varied between 20 and 60 min (see Table 3).

The names given to the near-crater measurement sites in the figures and in Table 3 indicate the day on which the measurements were taken and the order in which they were measured. For example, site “d2-27” was the second downwind site measured on 27 July. All upwind measurements were taken at the same location. In Table 3, the upwind site data sets collected on each day have been treated separately. For example, “up-30” is the set of measurements taken at the upwind site on 30 July.

2.2 Instrumentation

Ozone mixing ratios were measured via UV adsorption with a 2B Technologies instrument, model 202. A particle filter was placed on the inlet to prevent aerosols from influencing data or damaging the instrument. Because of the overlap of the O_3 and SO_2 adsorption spectra (e.g. Vance et al., 2010) we also fitted the inlet with two SO_2 scrubbers in series to prevent interference in the O_3 measurements. The scrubbers were packed with CrO_3 -coated microfibre filter paper. We tested the endurance of the scrubbers prior to the campaign. Air with a SO_2 mixing ratio of $700 \mu\text{mol mol}^{-1}$ was passed through the setup at a flow rate of 0.58 L min^{-1} . No SO_2 -related signal could be detected on the O_3 monitor for 2.25 h, a removal of approximately 15 mg of SO_2 . This greatly exceeds the expected SO_2 exposure of the scrubbers in the field.

New scrubbers and particulate filters were used daily. We calibrated the ozone monitor and inlet setup daily with a 2B Technologies model 306 O_3 calibration source. Linear losses of O_3 in the inlet/scrubber system was detected, the field results each day were corrected using the results from the calibration. We determined a measurement error of $\pm 3 \text{ nmol mol}^{-1}$ from the calibration results, which incorporates both random error of the instrument and uncertainties in the calibration correction.

We measured several gases' mixing ratios with a portable version of the INGV-type multi-component gas analyser system (MULTIGAS) (Aiuppa et al., 2011, 2012), though only the SO_2 measurements have been used in this analysis. The SO_2 measurement used an electrochemical sensor (0–200 $\mu\text{mol mol}^{-1}$; 3ST/F electrochemical sensor, City Technology Ltd.). A protective particulate filter was included on the instrument inlet. Data were recorded on an on-board data-logger and stored internally. The instrument's sensors and data-logger were housed in a water-proof box with a 1.2 L min^{-1} pump and a 12V battery. The sensors were calibrated before and after fieldwork with standard calibration gases; the instrument's accuracy and precision were determined to both be about 20%. The MULTIGAS was not used for the first 2 days of the campaign (24 and 25 July).

Temperature, humidity and wind vector were measured with a Kestrel 4500 commercial weather station, mounted on a tripod. Prior to the campaign, the device's relative humidity measurement was calibrated with standardised salt solutions.

Each instrument recorded data at a different frequency. Table 2 shows the instruments' measurement intervals and precision errors.

At the measurement sites we set up the ozone monitor and MULTIGAS on the same tripod, with the inlet tubes positioned closely together. The weather station tripod was positioned within 5 m of the tripod supporting the in situ instruments. The locations of the sites were determined by handheld GPS. The source of the plume at each downwind site was determined visually.

Remote sensing measurements of the downwind evolution of the volcanic plume with two Mini-MAXDOAS (multi-axis differential optical absorption spectroscopy) instruments were carried out (see Bobrowski et al., 2007, for an instrument description). Scans approximately perpendicular to the plume axis were made at distances between 6 and 16.6 km from the craters, corresponding to plume ages between about 8 and 30 min.

2.3 Data processing

The instruments listed in Table 2 each had independent internal clocks which could only be roughly synchronised in the field. The MULTIGAS' internal clock does not record absolute time, but only time since the device was switched on. In the field, we turned the device on at the same time as the other instruments, however it is probable some de-synchronisation of the data sets has occurred because of this. Inspection of the O_3 and SO_2 signal suggests that this de-synchronisation may have been up to 25 s.

We used the upwind site data sets as reference values of the background air. For each downwind site, the O_3 concentration time-series was converted to a time-series of O_3 change (ΔO_3) using the average O_3 mixing ratio measured at the upwind reference site on that day.

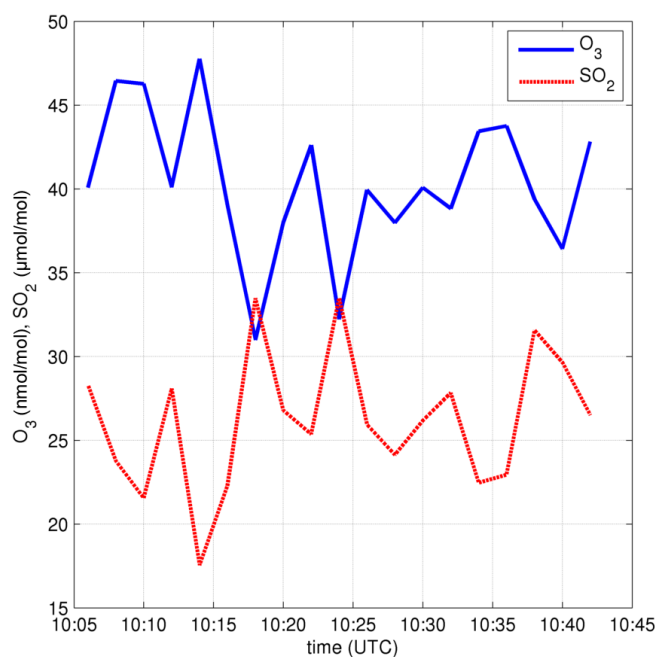


Figure 2. An example time series of measurements: O_3 and SO_2 mixing ratios measured in the plume at site d1-27. Data have been averaged into 120 s bins as described in the text.

As the collection frequency of the instruments differed, the data were binned into 120 s bins. Each 120 s bin therefore covered several measurements from each instrument and for each parameter a mean value was calculated. The use of 120 s bins reduced the precision error on the resulting data points. Additionally, the use of 120 s bins is expected to reduce the impact of any potential asynchronisation resulting from either the problem discussed above or from different instrument response times.

Inspection of the signals (e.g. Fig. 2) showed clear anticorrelation between the O_3 and SO_2 mixing ratios.

SO_2 is used as the inert plume tracer, a measure of plume strength for the bin. Measured background levels of SO_2 were negligible in comparison to in-plume measurements, and SO_2 is expected to be effectively chemically inert. For each 120 s measurement bin a ratio $\Delta\text{O}_3 / \text{SO}_2$ (change in molecules O_3 per molecule inert plume tracer) was calculated, this is a measure of O_3 change that is attributable to chemical and not physical processes (a proof of this can be found in the appendix).

Transport times between the craters and the downwind sites were determined from the distances between the sites and the vents, and the measured wind speeds.

For each bin, the uncertainty in the wind speed is calculated by adding in quadrature the expected error of the Kestrel instrument and the variance of the individual wind-speed measurements which are averaged to yield the wind speed value for the bin. This error can be seen in the x axis error bars on Figs. 3, 4, and 5. Similarly the uncertainty in

Table 2. Instruments used at the near crater sites.

Name	Parameter	Precision	Sampling interval
Kestel 4500	Temperature	± 1 °C	30 s
	Humidity	± 3 % RH	
	Wind speed	± 3 % or ± 1 m s ⁻¹ (whichever larger)	
MULTIGAS	SO ₂	± 20 %	2 s
2B Technologies Model 202	O ₃	± 3 nmol mol ⁻¹	10 s

the SO₂ values is calculated from the variance of this quantity within the bin and the instrument error. The uncertainty in the O₃ values is calculated in the same manner, and for Δ O₃ measurement additionally factor in the variance in the O₃ measurement in the corresponding background measurement. These errors are manifest in the y axis error bars of Figs. 3 and 4, and are combined to give the y axis error bars of Fig. 5.

Gaseous elemental mercury is known to be emitted by Etna (Bagnato et al., 2007) and has the potential to cause interference with UV-based ozone monitors (EPA, 1999), giving false O₃ signal when in the sampled air. The inlet and CrO₃ scrubbers were not tested against mercury interference. If elemental mercury was entering the monitor in the mixing ratios observed by Bagnato et al. (2007), and causing interference of the magnitudes seen by EPA (1999), it would have caused us to make underestimations of the depletion of O₃ on the order of 10 nmol mol⁻¹. However, this would assume that all the mercury emitted from the volcano was primarily in the elemental form. This is unlikely as modelling studies (von Glasow, 2010) indicated that much on the mercury is oxidised within the plume. In addition, mercury measurements at Etna contemporaneous with our campaign (T. A. Mather and M. L. Witt, personal communication, 2014) indicate that the mercury loading of the craters' plumes were significantly less than that which was measured by Bagnato et al. (2007) in 2004–2007 and confirmed that much of the observed mercury was in an oxidised or particulate form. It is also probable that the two CrO₃ scrubbers scavenged or oxidised much of the elemental mercury in the sampled air, as the system is not heated and the scrubbers are changed daily it is unlikely that mercury scavenged by the scrubber would be re-volatilised as elemental mercury. No compensation for this potential effect was made in the post-processing as we believe that the potential for this interference is very low and because the observed anti-correlation of the O₃ and SO₂ signals is the inverse of what would be expected from such interference.

The MAXDOAS data were evaluated for BrO and SO₂ with the WinDOAS V2.10 software package (van Roozendaal and Fayt, 2001) which uses a nonlinear least-square method (Stutz and Platt, 1996). A BrO/SO₂ ratio was calculated for each measurement day by a correlation plot of the daily data set. The plume ages at the point of measurement were calculated from the wind speed and lateral distance

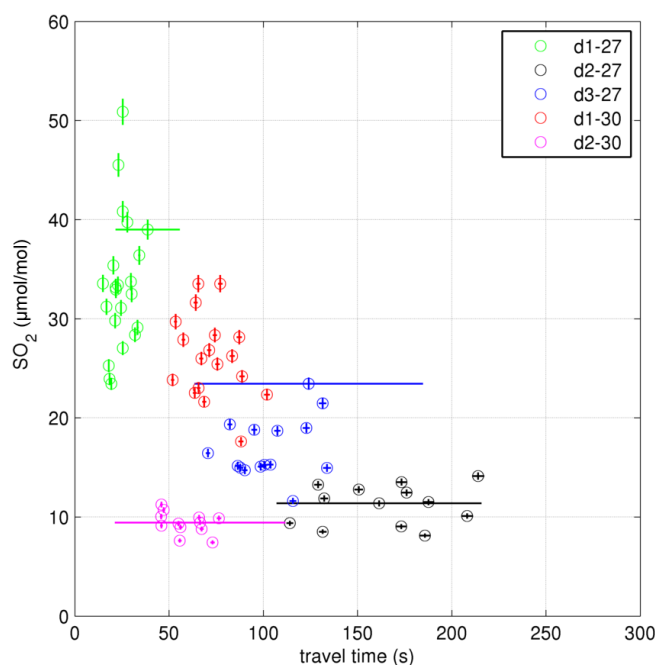


Figure 3. SO₂ mixing ratios against calculated processing time at the in-plume measurement sites. Error bars are 1 standard deviation, accounting for both variance in the values over the 120 s of each bin and known errors of the instruments.

from the craters. In the absence of wind speed data at Etna, speeds from soundings taken at Trapani (37.91° N, 12.50° E, data from <http://weather.uwyo.edu/upperair/sounding.html>) were used. The 220 km distance between the location of the soundings and the volcano means that there is a relatively large and unquantifiable uncertainty in the calculated plume ages.

3 Results

An overview of the near-crater results from this campaign can be seen in Table 3. Calculated plume ages were found to be of the order 1–4 min.

The plume was observed to be grounded at all downwind sites, enabling measurement. As both O₃ and SO₂ are known to exhibit deposition to the ground, there is the potential for the mixing ratios of these species at the elevation of the in-

Table 3. Overview of results from Etna summit measurement campaign.

Site	Upwind/ Downwind	Plume Source	Distance to source/m	Date	Time (UTC)	Average Values					
						Temp./ °C	Humidity/ %	[SO ₂]/ μmol mol ⁻¹	[O ₃]/ nmol mol ⁻¹	Δ[O ₃]/ nmol mol ⁻¹	
up-24	Upwind	–	–	2012-07-24	09:10–09:35	3	49	3	n.m.	86	–
d1-24	Downwind	Central	300	2012-07-24	10:05–11:30	7	64	7	n.m.	69	–18
up-25	Upwind	–	–	2012-07-25	11:05–11:30	7	63	7	n.m.	83	–
d1-25	Downwind	Central	440	2012-07-25	09:52–10:30	11	60	11	n.m.	58	–25
up-27	Upwind	–	–	2012-07-27	08:41–09:02	9	59	9	0	77	–
d1-27	Downwind	NEC	150	2012-07-27	09:26–10:16	11	49	11	34	46	–31
d2-27	Downwind	NEC	380	2012-07-27	10:34–11:06	13	42	13	11	50	–26
d3-27	Downwind	NEC	240	2012-07-27	11:20–11:50	13	34	13	17	41	–35
up-30	Upwind	–	–	2012-07-30	09:12–09:34	12	<5	12	0	73	–
d1-30	Downwind	Central	240	2012-07-30	10:05–10:43	16	6	16	26	40	–32
d2-30	Downwind	Central	300	2012-07-30	11:07–11:41	15	5	15	9	62	–11

n.m.: SO₂ was not measured on these days.

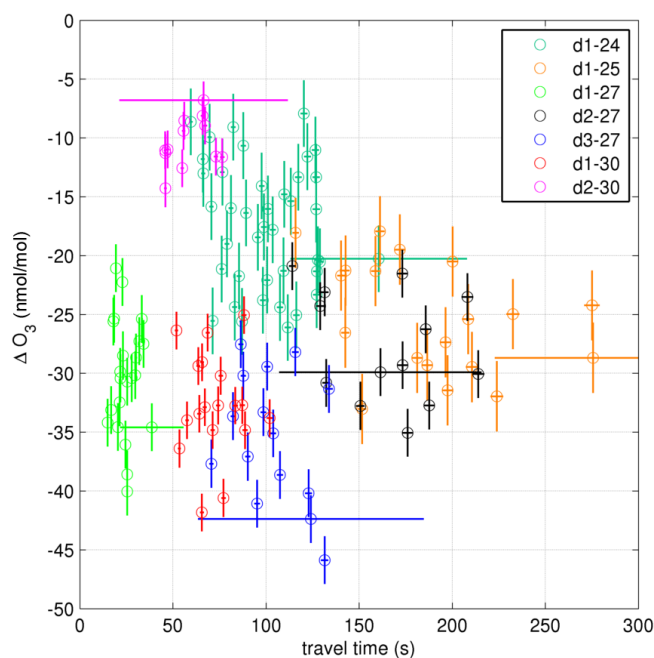


Figure 4. ΔO₃ against calculated processing time at the in-plume measurement sites. Error bars are 1 standard deviation, accounting for both variance in the values over the 120 s of each bin and known errors of the instruments.

struments to be perturbed by this physical process. However the ground at the peak of Etna is rocky and devoid of vegetation and could be classed in the scheme of Wesely (1989) as “barren land”. The bulk surface resistance of this category sufficiently high that, even if the aerodynamic and sublayer resistances are low, the deposition velocity will be too low to cause a significant perturbation during the transport from crater rim to the measurement sites.

Ozone depletion of up to 45 % was observed at all seven downwind sites. SO₂ decreases with transport time (Fig. 3), as would be expected due to dilution. The low SO₂ mixing

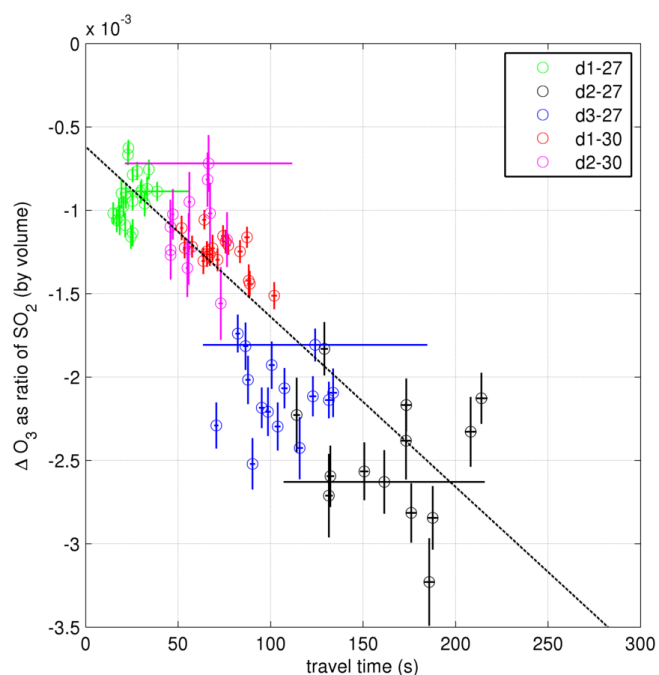


Figure 5. ΔO₃ / SO₂ against calculated processing time at the in-plume measurement sites. The gradient of the linear line of best fit is $-(1.02 \pm 0.07) \times 10^{-5} \text{ s}^{-1}$ and the y-intercept is $(-6.2 \pm 0.5) \times 10^{-4}$. Error bars are 1 standard deviation, accounting for both variance in the values over the 120 s of each bin and known errors of the instruments.

ratios at site d2-30 suggest this site was at the plume edge, this is consistent with visual observations of the plume. A plot of ΔO₃ against transport time (Fig. 4) shows no comparable trend as chemical destruction of O₃ and dilution with background air occur simultaneously.

In Fig. 5, Δ[O₃] / [SO₂] vs. transport time is plotted. The linear trend is indicative of chemical destruction of O₃ occurring within the first few minutes of plume transport (see Appendix).

The data on Fig. 5 appear to fit a linear trend. This indicates, that as a ratio to the inert plume tracer, the rate of chemical O₃ destruction is constant within the first few minutes of plume evolution. In the expected chemical cycles Eq. (2), the rate of O₃ destruction is dependant on the concentration of halogens in highly reactive (non-hydrogen halide) forms. Therefore a constant rate of reaction indicates that the ratio of these species to SO₂ is approximately constant during the first few minutes of plume evolution.

Depletion of ozone occurred on 30 July even when the relative humidity was <10%. This is below the crystallisation point of any major aerosol component that would be expected to be present in the plume. It would therefore be expected that at such low humidities aqueous phase reactions (such as reaction 1d) would occur at a much slower rate, preventing the bromine explosion from converting HBr to very reactive forms of bromine. On inspection of the humidity measurements (Table 3), no significant increase in humidity due to volcanic H₂O was observed at the downwind sites (the humidity variation between up-24 and d1-24 could be volcanic in origin, however it is also in the range of normal meteorological variability). The possible impacts of a short-lived elevated humidity within the crater due to volcanogenic H₂O are considered with the “dry” model runs in Section 4 of this paper.

Data points from all measurement sites appear to approximately fit the same line-of-best fit on Fig. 5. This is despite the measurements spanning 2 separate days and these 2 days having very different circumstances. On 27 July the measurements were made in the plume of the North-East crater, and on 30 July the measurements were made in the plume of the central craters. The different craters are known to exhibit differing emission compositions (Aiuppa et al., 2005a). In addition, the relative humidity on these 2 days was markedly different. It is unknown whether the cause of these two data sets’ approximate alignment in Fig. 5 is due to mechanistic reasons relating to the “effective source region” (see below) or coincidence.

A linear line of best fit was calculated using a Williamson–York iterative bivariate analysis (Cantrell, 2008), its gradient is $(-1.02 \pm 0.07) \times 10^{-4} \text{ s}^{-1}$. The r^2 value of this fit is 0.76. This linear trend can be interpreted as there being a rate of destruction of O₃ within the plume of approximately 10^{-4} molecules per molecule of SO₂ (the inert plume tracer) per second in the first few minutes of plume evolution. SO₂ is only used as a measure of plume intensity, SO₂ does not play a role in O₃ destruction.

Extrapolating the line of best fit on Fig. 5 results in an intercept of the x axis of approximately -60 s. Assuming chemical O₃ destruction is not instantaneous in the plume, it would be expected that the line of best fit would pass through the origin. We determine that this offset is most likely the result of inaccuracy in the calculation of processing time. Firstly, the calculation assumes that air travels in a straight line between vent and sampling point; however, the

path length is likely to be greater than this due to turbulence, the topology of the mountaintop, and variation in wind direction. Second, the distances between vents and sampling points were calculated only from their lateral displacement and did not consider the time taken for vertical transport from the vent to the crater rim. From our visual observations of the plume, we estimate this vertical transport time is likely to have been of the order of a minute.

Table 4 lists the composition of volcanic volatiles. These figures are calculated from June 2012 values of SO₂, HF, HCl, and HBr measured using various alkaline traps (Wittmer et al., 2014) (Bocca Nuova crater), as well as the MULTIGAS measurements of CO₂ and SO₂ made on the main campaign (using all available data). As complete characterisation of the volcanic volatiles could not be made from the June 2012 measurements alone, the ratio of the gases to CO₂ were used in these calculations with absolute values of H₂O and CO₂ calculated from the values reported in (Aiuppa et al., 2005b). Additionally, H₂ / SO₂ was taken from (Aiuppa et al., 2011), and H₂S / SO₂ was calculated from the values reported in (Aiuppa et al., 2005b). These values were then normalised to ensure that the sum equals unity.

The remote sensing results are shown in Table 5, along with the correlation between the BrO and SO₂ columns and the estimated plume age (based upon distance from the crater and wind speed, see Sect. 2.3). The BrO / SO₂ column ratios are within the range typically measured both in the Etna plume (Bobrowski and Giuffrida, 2012) and volcanic plumes in general (Boichu et al., 2011; Hörmann et al., 2012), though they are at the lower end of these ranges.

4 Modelling study

We used a vertically resolved 1-D numerical model MISTRA to simulate the chemical evolution of the volcanic plume in the atmosphere and to test the model’s performance against the data presented above. The model is the same as used in von Glasow (2010) with minor modifications. Based on previous work (Gerlach, 2004; Martin et al., 2006; von Glasow, 2010) we used the thermodynamic equilibrium model HSC (Roine, 2007) to calculate the composition of the mixture of volcanic volatiles and ambient air in the “effective source region” which is located in the crater where temperatures are high enough so that the assumption of immediate thermodynamic equilibrium is still largely valid but also enough ambient air has been entrained to oxidise the main volcanic reduced gases (mainly H₂S and H₂) which leads to a dramatic change in halogen speciation. Recent studies (e.g. Aiuppa et al., 2007; Martin et al., 2012) have shown that the assumption of thermodynamic equilibrium is incorrect for a number of compounds, especially for H₂S and NO_x which makes the use of an equilibrium model such as HSC problematic but it is still regarded to provide a better approximation of the composition of volcanic volatiles than not assuming any equi-

Table 4. Fractional composition (by volume) of pure volcanic volatiles of Mt. Etna in June 2012 (“low Br/S”) and in 2005 (“high Br/S”). These compositions were used to initialise the HSC calculations (for results see Table S1). “High Br/S” refers to the plume composition used in von Glasow (2010).

Species	low Br / S	high Br / S
H ₂ O	8.75×10^{-1}	8.57×10^{-1}
CO ₂	9.76×10^{-2}	9.56×10^{-2}
SO ₂	1.63×10^{-2}	2.86×10^{-2}
H ₂ S	8.14×10^{-4}	1.43×10^{-3}
H ₂	2.11×10^{-4}	3.72×10^{-4}
HF	2.96×10^{-3}	2.20×10^{-3}
HCl	6.87×10^{-3}	1.43×10^{-2}
HBr	9.99×10^{-6}	7.04×10^{-5}

bration. See also related comments in Roberts et al. (2014). Compared to von Glasow (2010) we used a higher ratio of volcanic volatiles : ambient air (95 : 5), which is very rich in volcanic volatiles but already oxidised.

The model is run in Lagrangian mode, following the evolution of a “puff” of plume downwind of the volcano and explicitly modelling the interaction of the plume with background air (see von Glasow, 2010). The modelled wind speed is 10 m s^{-1} , and the plume dispersal is assumed to be Gaussian and follows Kärcher (1999) using the Pasquill–Gifford scheme to calculate the parameters of the Gaussian plume (see e.g. Seinfeld and Pandis, 2006).

The S and Br elemental ratios of the volcanic volatiles used in the HSC calculation are taken from the average values measured at the crater rim (Table 4, column “low Br / S”) on 26 and 27 June 2012. We also did a second set of model runs initialised with a much higher Br / S ratio (Table 4, column “high Br / S”) based on unpublished data from Alessandro Aiuppa. Table S1 in the Supplement gives the composition of the volcanic volatiles and the results of the HSC calculations used as input for the 1-D model.

The relative humidity varied significantly during the campaign (see Table 3), which is important as the bromine explosion is efficient in the presence of aqueous aerosol.

Following Seinfeld and Pandis (2006) a crystallisation humidity of 40 % for sulphate-containing particles is assumed in the model, but other studies suggest sulphate-containing particles to be liquid for lower relative humidities (e.g. Martin et al., 2003).

Regardless of the exact composition and crystallisation humidity of the volcanic aerosol present on the days of this study when simultaneous SO₂ and O₃ measurements were made, we wanted to test the implications of crystallised aerosol in the model. We therefore ran the model at the following three different settings regarding the deliquescence of the aerosol:

1. “moist”: relative humidity $\sim 57\%$ (above the sulphate crystallisation humidity assumed in the model); aerosol aqueous phase chemistry occurs throughout the plume;
2. “dry”: relative humidity $\sim 21\%$ (below the assumed sulphate crystallisation humidity); aerosol aqueous phase chemistry consequently disabled; exception is the first ~ 1 min, where water from the volcano elevates the humidity within the plume;
3. “noHet”: where all heterogeneous chemical reactions were deactivated.

Being remote from major sources of anthropogenic pollution and above the planetary boundary layer, the ambient NO_x levels at Etna are expected to be negligible compared to the amount of nitrogen oxides (NO_x) that thermodynamic models such as HSC predict would be formed in the effective source regions of volcanic vents. However Kaschka (2007) and Martin et al. (2012) discussed the problems associated with this, mainly that the time required to reach equilibrium for NO_x under the given temperatures in the effective source region is, by far, too long. The generation of NO_x could significantly perturb bromine chemistry due to the formation of BrNO₂ (see e.g. Fig. 2 in von Glasow, 2010). A detailed discussion of volcanic plume Br–NO_x chemistry, including a modelling study that includes additional BrNO₂ reactions, can be found in Roberts et al. (2014). Use of their scheme would likely result in less bromine being present as BrNO₂ and might lead to somewhat different O₃ development. For each of the three aerosol settings described above, two model runs were done – one with volcanic NO_x in the quantities predicted in the thermodynamic model and one with no volcanic NO_x. These six runs were performed for both of the plume initialisations shown in Table 4.

4.1 “Low Br / S” modelled plume

Model results are shown for the core of the plume which is most appropriate for the grounded plume measured in our campaign. Comparison of modelled and measured plume SO₂ mixing ratios, shown in Figs. 6 and 3, respectively, show the plumes to be of similar intensity. Differences in SO₂ mixing ratio between the six model runs are negligible. In order to approximately reproduce the measured SO₂ mixing ratios at the crater rim, the “initial dilution” ratio of the HSC output with ambient air (see Materials and Methods in von Glasow, 2010) was set to a factor of $9\times$.

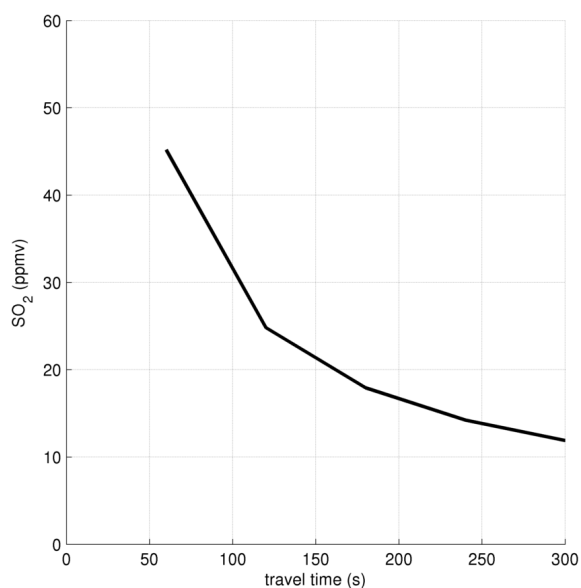
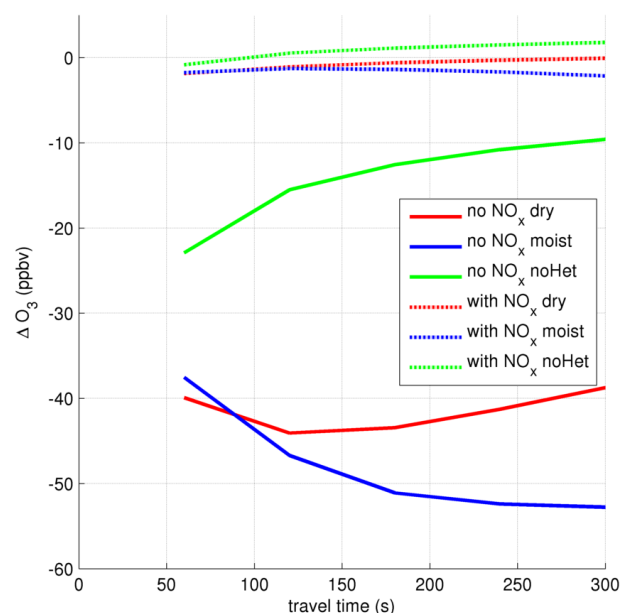
The 0–60 s section of the following figures are not plotted as the first data output from the model occurs at 60 s after plume release.

Figure 7 shows the modelled ozone depletion in the six runs. Figure 8 shows this as a ratio to SO₂ mixing ratio (c.f. Figs. 4 and 5, respectively).

In comparison to the measured data where bromine chemistry at the craters can only be inferred from O₃ and SO₂ data,

Table 5. Spectroscopic measurements of the Mount Etna plume. These data are also plotted on Fig. 11. R^2 refers to the correlation of BrO and SO₂ within the measurements.

Date	BrO / SO ₂	R^2	Distance (km)	Wind speed (m s ⁻¹)	Plume age (min)
2012-07-24	0.9×10^{-4}	0.94	11	8.7	21
2012-07-24	0.7×10^{-4}	0.80	6.3	8.7	12
2012-07-25	1.2×10^{-4}	0.66	10.4	6.2	28
2012-07-25	1.3×10^{-4}	0.52	6.8	6.2	18
2012-07-26	1.1×10^{-4}	0.64	17	9.3	30
2012-07-26	1.0×10^{-4}	0.90	8.8	9.3	16
2012-07-27	1.2×10^{-4}	0.48	16	22	12
2012-07-28	0.8×10^{-4}	0.47	12	13	15
2012-07-28	0.9×10^{-4}	0.50	9.0	13	11
2012-07-31	0.6×10^{-4}	0.72	5.9	5.7	17
2012-08-02	1.0×10^{-4}	0.48	6.0	12	8.1

**Figure 6.** Modelled SO₂ mixing ratio in the plume core (“low Br / S” initialisation).**Figure 7.** Evolution of ΔO_3 in the core of the plume for the six model scenarios discussed in the text (“low Br / S” initialisation).

the modelled bromine chemistry can be investigated directly; Fig. 9 shows the speciation of bromine in the six different runs.

In all three runs with volcanic NO_x, the bromine almost entirely partitions to BrNO₂. BrNO₂ does not contribute to O₃ depletion. As shown in Fig. 7, in these runs the net O₃ change is almost zero for all runs and is actually positive in the “noHet” and “dry” cases as a consequence of O₃-generating NO_x chemistry.

The bromine explosion, as evidenced by conversion of HBr to other forms of bromine can be seen in the “moist” runs and has also occurred between 0 and 60 s of the “dry” run (where volcanic H₂O is modelled to increase the plume’s

humidity). In the “noHet” runs and in the “dry” runs after the volcanic humidity increase has ended, the processes that convert HBr to other forms are effectively halted.

In all three NO_x-free runs, O₃ is depleted. Reactive bromine is generated by the high-temperature processes of the effective source region in all three runs. In the “moist” case, and to a slightly reduced extent in the “dry” case, this is supplemented by the generation of reactive bromine species via the bromine explosion, to the extent that in the “moist” case the O₃ depletion is near 100%. The consequences of these three runs’ differences in bromine chemistry can be

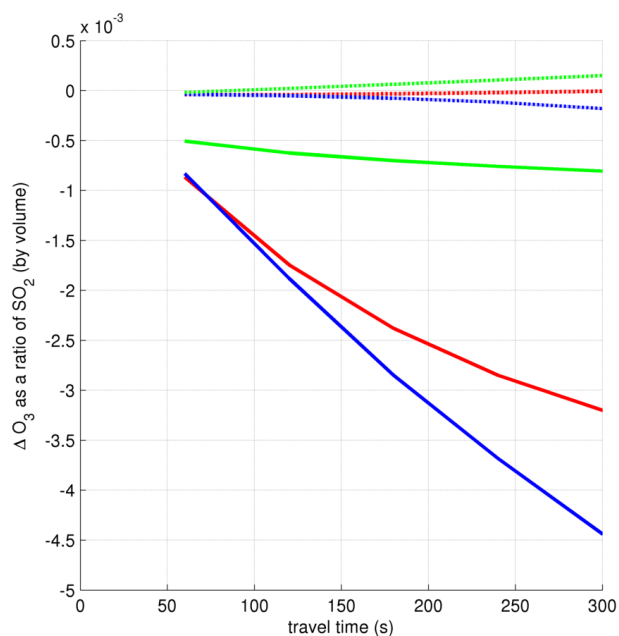


Figure 8. Evolution of $\Delta O_3 / SO_2$ in the core of the plume for the six model scenarios discussed in the text (“low Br / S” initialisation). The colour code is the same as in Fig. 7.

seen on Fig. 8, where the $\Delta[O_3] / [SO_2]$ ratios diverge over the first few minutes.

A comparison of the O_3 depletion (Figs. 7 and 8) after 60 s for the “noHet” run with that for the “dry” and “moist” runs shows the comparative influence of different processes within the crater. The ozone depletion in the “noHet” case must be due to reactive halogen generated in high temperature processes as heterogeneous reactions are required for low-temperature reactive bromine generation (the bromine explosion). The additional ozone depletion at 60 s in the “dry” and “moist” cases can therefore be attributed to the bromine released via the bromine explosion within the crater. As the plume moves away from the crater, the heterogeneous bromine explosion can only occur in the “moist” run so the increasing differences between the “moist” and “dry” runs are due to the continued bromine explosion in the “moist” run.

The model results can also be used to calculate vertical columns of BrO and SO_2 (Fig. 10) and the corresponding ratios (Fig. 11). These figures show these data over the first 60 min of plume evolution and are also overlaid with the spectroscopic observations listed in Table 5. Due to logistical constraints, we have concurrent near-crater and downwind data only for 27 July; this datapoint is highlighted on Fig. 11. As can also be seen in Fig. 9, only the “moist” runs maintain significant amounts of volcanic bromine as BrO and show column BrO / SO_2 ratios comparable with measurements. Also notable is the rise in the column ratio in the “moist-with NO_x ” run after a few minutes. After signifi-

cant dilution of the plume has occurred, the dominance of the Br NO_2 reservoir diminishes. Thus volcanic NO_x , if present, delays but does not prevent in-plume reactive bromine chemistry.

4.2 “High Br / S” modelled plume

The elemental emissions ratio of Etna is known to vary between craters and over time (Aiuppa et al., 2005a). Figure 12 shows the evolution of modelled ΔO_3 for the six runs where they have been initialised using the “high Br / S” ratios (see Table 4). Further plots relating to these are shown in the supplemental material to this paper (Figs. S1–S5 in the Supplement). As for the runs described previously, in order to approximately reproduce the measured SO_2 mixing ratios at the crater rim (and those from the prior runs) an “initial dilution” ratio was again used, in these runs it was set to a factor of $15\times$ dilution.

The increase in the amount of volcanic bromine has a significant impact on the chemistry. In the NO_x -free runs the reactive bromine generated by the effective source region is sufficient to destroy almost all of the O_3 within the plume and the plume chemistry is O_3 -limited. As can be seen in Table 1 of the supplementary material, the NO_x / S ratio is very similar for the “low Br / S” and “high Br / S” cases. As a consequence the Br/ NO_x is very different, being 0.0875 in the “low Br / S” runs and 3.35 in the “high Br / S” runs. Whereas in the “low Br / S” runs with volcanic NO_x almost all reactive Br had partitioned to Br NO_2 , in the corresponding “high Br / S” the partitioning of the reactive bromine to Br NO_2 is more moderate and a significant amount of reactive bromine in other forms remains. This results in moderate O_3 depletion in the early plume, with similar variation between the humidity cases as discussed above.

4.3 Comparison with measured data

This modelling study shows that the plume chemistry is highly sensitive to several factors: the presence of volcanic NO_x , the magnitude and composition of the primary volcanic emissions, and heterogeneous processes which are dependant on aerosol and humidity. Any comparison between the model results and the measurements must be made with caution. The following points should be stressed:

- Volcanic volatile Br / S ratios are known to vary considerably both over time and between Etna’s different craters (Aiuppa et al., 2005a). The composition of the volcanic volatiles used in the “low Br / S” modelling study was based on measurements which were taken a month prior to the O_3 and SO_2 measurements. It is unknown how the emissions may have changed in the intervening period, and comparing the results with the two different initialisations shows that variations in volatile composition can be very significant.

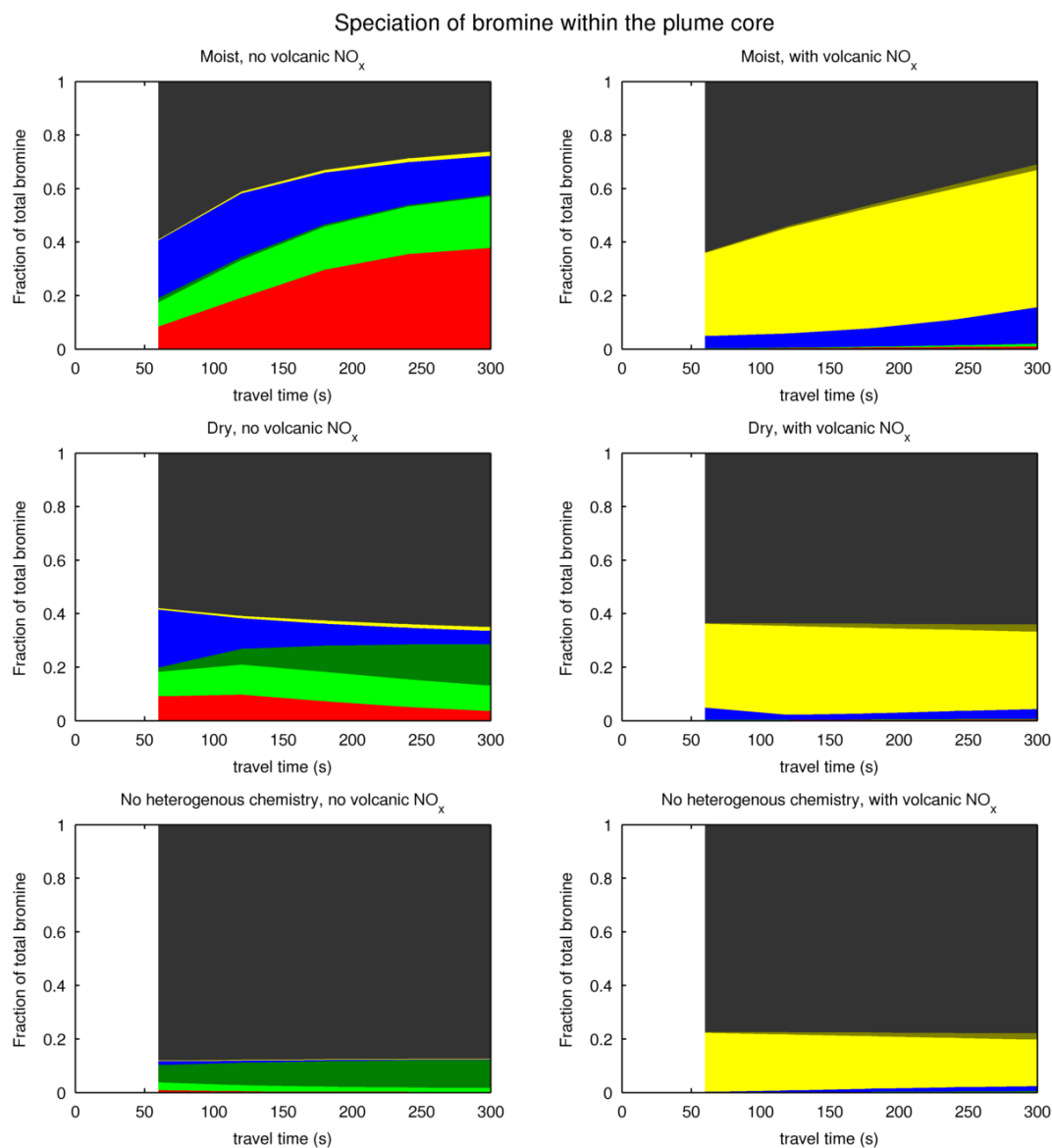


Figure 9. Speciation of bromine for the six model scenarios discussed in the text (“low Br / S” initialisation). Red – Br; light green – BrO; dark green - HOBr; blue – Br₂; yellow – BrNO₂; grey – HBr.

- Over the time that measurements were taken, the magnitude and composition of the volcanic emissions may have varied significantly. This cannot be reproduced in the model.
- We focussed on model output from the plume core. However, the varying wind velocity at the measurement sites mean that sampling will have been from varying parts of the plume.

As such it is not possible to conclude definitively from the comparison of the measurement and modelling studies the exact chemical processes that were occurring within the plume at the time of the measurement campaign. Nevertheless, the modelling study shows that the reactive bromine

generated in the effective-source region is sufficient to cause appreciable ozone depletion within the early plume, however the effect is stronger and more sustained when further reactive bromine can be generated by the bromine explosion.

The model runs which are the best fit for the near-crater O₃ and SO₂ measurements are not the same runs which best match the BrO / SO₂ columns measured further downwind. This is a discrepancy which highlights the need for further investigation into plume chemistry and the relationship between in-plume bromine, NO_x and humidity.

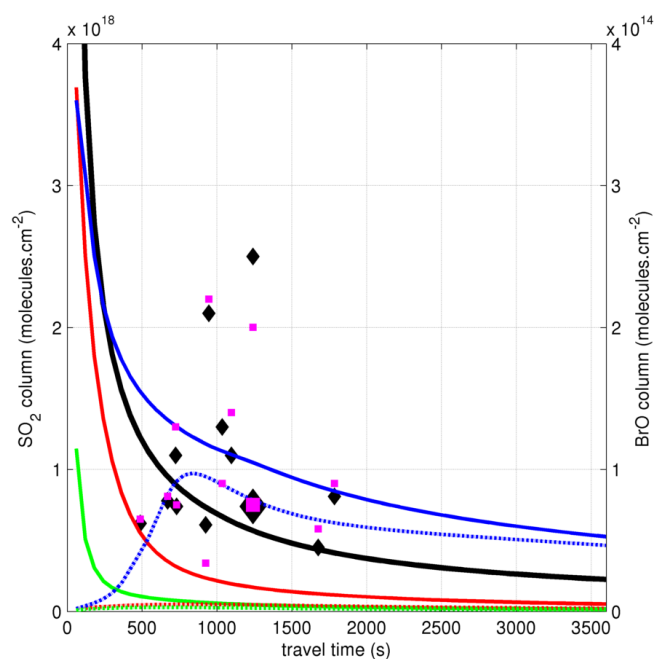


Figure 10. Evolution of vertical column SO_2 and BrO for the six model scenarios discussed in the text (“low Br / S” initialisation). SO_2 is plotted against the left axis scale in black (variation between the scenarios is minimal). The BrO columns are plotted against the right axis scale using the same colour code as in Fig. 7. The black diamonds represent the spectroscopic measurements of SO_2 , the magenta squares represent the spectroscopic measurements of BrO. The larger markers represent the 2012-07-27 measurement contemporaneous with near-crater measurements of both O_3 and SO_2 .

5 Conclusions

In this study of chemical evolution in the plumes of Mount Etna during a passively degassing phase we have confirmed O_3 depletion to occur less than a few minutes after emission. This extends the limited data set regarding volcanic O_3 depletion and is the first data set to date based on ground-based data to include simultaneous measurements of O_3 and a plume tracer. The depletion process is ongoing over this time period, and the data suggests it occurred at an approximately constant rate. For the Etna craters measured in late July 2012, this rate was found to be a depletion of approximately 10^{-4} molecules of O_3 per molecule of SO_2 (the inert plume tracer) per second.

Substantial depletion of O_3 was observed within minutes after emission, and occurred at very low humidities at which the heterogeneous processes of the bromine explosion would be significantly slowed. On the basis of this, an approximately constant rate of ozone depletion (as a ratio of the inert tracer), and analysis of our modelling study we conclude that most of the very reactive halogens that caused the observed O_3 depletion within the 1–4 min old plume were not a product of chemical cycling in the plume once

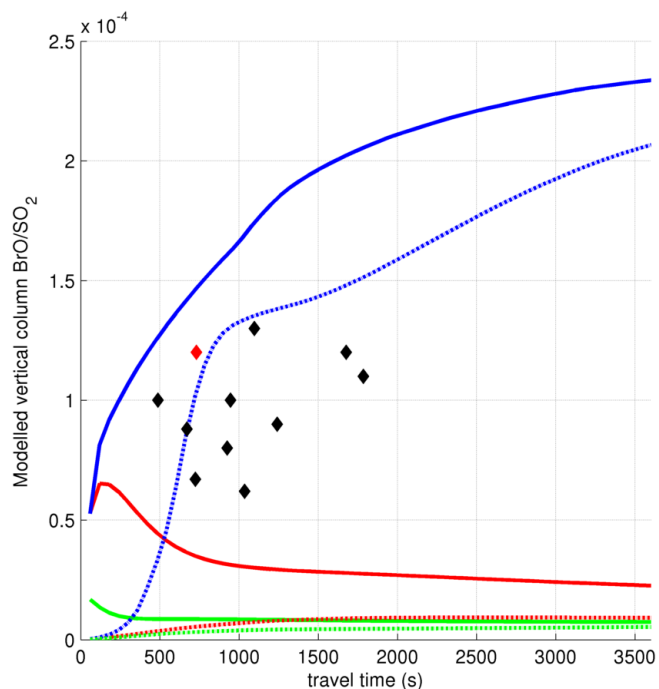


Figure 11. Evolution of vertical column BrO / SO_2 ratios for the six model scenarios discussed in the text (“low Br / S” initialisation). The colour code is the same as in Fig. 7. The black diamonds represent the spectroscopic measurements listed in Table 5, the red diamond represents the 2012-07-27 measurement contemporaneous with near-crater measurements of both O_3 and SO_2 .

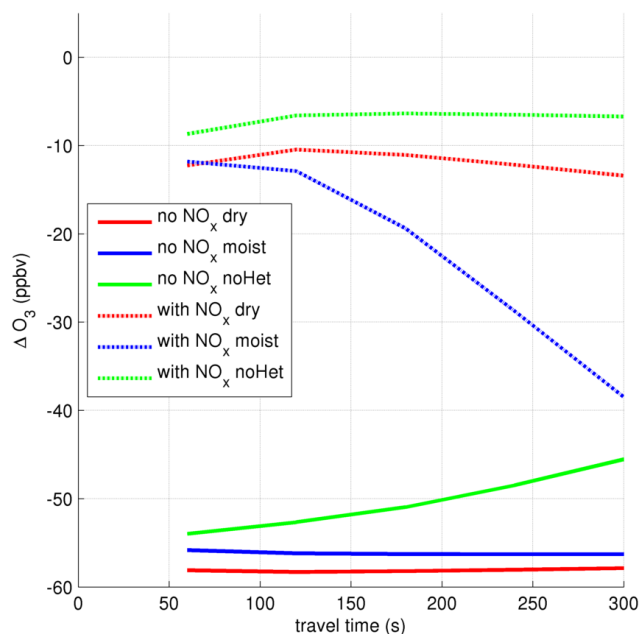


Figure 12. Evolution of ΔO_3 in the core of the plume for the six model scenarios discussed in the text using the “high Br / S” initialisation. The colour code is the same as in Fig. 7.

it had emerged from the crater but rather generated in *effective source regions* near the vents in high-temperature processes that ceased shortly after emission, and/or very fast low-temperature chemical processes within the crater. This is in agreement with the thermodynamic modelling studies of Gerlach (2004), Martin et al. (2009), and von Glasow (2010) which predict oxidised halogens being produced by high-temperature effective source regions in volcanoes. The in situ measurements in this study were taken at distances less than 500 m from the vents. Measurements of BrO / SO₂ at such distances tend to be low compared to those measured further downwind (Bobrowski et al., 2007; Boichu et al., 2011). That substantial O₃ depletion can occur in such regions suggests these low levels of reactive bromine (and possibly chlorine) species that are generated within the crater are sufficient to cause the O₃ depletion observed in this study.

We have also shown that further downwind, contemporaneous spectroscopic observation of the plume shows BrO / SO₂ columns are within the typical range that has been observed within volcanic plumes. Our near-crater measurements are therefore likely to be typical for Mount Etna in its passively degassing phase.

Results from our modelling study suggest that a significant factor influencing the bromine-ozone chemistry in the very early plume is the presence or absence of volcanic NO_x and, if it is present, its source strength relative to that of Br. The modelling study shows that volcanic NO_x would be expected to significantly retard early O₃ depletion and therefore plume O₃ and Br data could be compared to model output to yield information on NO_x within a plume. The measured data from this campaign are insufficient to definitively ascertain whether or not volcanic NO_x generation is occurring but it appears unlikely to be the case as near-crater O₃ depletion was observed whilst volcanic NO_x emissions suppressed such O₃ depletion in the modelling study. The modelling study also shows how variations in humidity, and the consequent variations in heterogeneous chemistry, may impact the plume chemistry in detectable ways. A comparison of the results from our three different humidity/heterogeneous chemistry cases gives an indication as to the extents that O₃ depletion can be ascribed to reactive halogens generated in high-temperature processes within the crater, low-temperature processes within the crater, and low temperature processes further downwind.

Appendix A: Proof that $\Delta O_3 / SO_2$ is a measure of chemical change only.

The following is a proof that $\Delta A / \Delta B$ cannot be changed by physical plume mixing processes, and thus any change observed in the ratio is the result of chemical change only.

Starting with a generic plume dispersal equation for a change in concentration due to mixing (von Glasow et al., 2003),

$$\begin{aligned} \left. \frac{dc}{dt} \right|_{\text{mix}} &= -\frac{1}{A_p} \frac{A_p}{dt} (c - c_{\text{bg}}) \\ &= -\omega(c - c_{\text{bg}}), \end{aligned}$$

where A_p is the cross-sectional area of the plume, $\omega = \omega(t)$ is the mixing parameter, c and c_{bg} are the concentration of the species in the plume and the background air, respectively.

Discretising,

$$\left. \frac{\delta c}{\delta t} \right|_{\text{mix}} = -\omega(c - c_{\text{bg}}).$$

Let c_i be the concentration of the species at time t_i :

$$c_2 = c_1 + \frac{\delta c}{\delta t} \delta t = c_1 - \omega(c_1 - c_{\text{bg}}) \delta t.$$

We define

$$\Delta c_i = c_i - c_{\text{bg}}.$$

Therefore,

$$\begin{aligned} \Delta c_2 &= c_1 - \omega(c_1 - c_{\text{bg}}) \delta t - c_{\text{bg}} \\ &= \Delta c_1 - \Delta c_1 \omega \delta t = \Delta c_1 (1 - \omega \delta t). \end{aligned}$$

Taking the ratio of gas A and gas B,

$$\frac{\Delta c_2^A}{\Delta c_2^B} = \frac{\Delta c_1^A (1 - \omega \delta t)}{\Delta c_1^B (1 - \omega \delta t)} = \frac{\Delta c_1^A}{\Delta c_1^B}.$$

In the volcanic case as in this study, $A = O_3$ and $B = SO_2$. The concentration of SO_2 outside of the plume is negligible in comparison to in-plume measurements, $SO_{2\text{bg}} \approx 0$ thus $\Delta SO_2 = SO_2$ and $\Delta O_3 / \Delta SO_2 = \Delta O_3 / SO_2$. Therefore,

$$\frac{\Delta O_{3,1}}{SO_{2,1}} = \frac{\Delta O_{3,2}}{SO_{2,2}}.$$

Therefore, if no chemical change in O_3 occurs between time 1 and time 2, the ratio remains unchanged.

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