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## D/H diffusion in serpentine

Hélène Pilorgé (1), Bruno Reynard (1), Laurent Remusat (2), Sylvie Le Floch (3), Gilles Montagnac (1), and Hervé Cardon (1)

(1) Laboratoire de Géologie de Lyon (LGL-TPE) - UMR CNRS 5276, Université de Lyon - ENS de Lyon - Université Lyon 1, F-69622 Villeurbanne, France (helene.pilorge@ens-lyon.fr), (2) Institut de Minéralogie, de Physique des Matériaux et de Cosmochimie (IMPMC) - UMR CNRS 7590, Sorbonne Universités - UPMC - IRD - Muséum National d'Histoire Naturelle, 57 rue Cuvier, Case 52, 75231 Paris Cedex 5, France, (3) Institut Lumière Matière (ILM) - UMR 5306, Université de Lyon - Université Lyon 1, F-69622 Villeurbanne, France

Interactions between aqueous fluids and ultrabasic rocks are essential processes in a broad range of contexts including hydrothermal alteration on the parent body of carbonaceous chondrites, at mid-oceanic ridge, and in subduction zones. Tracking these processes and understanding reaction kinetics require knowledge of the diffusion of water in rocks, and of isotope fractionation in major minerals forming under hydrous conditions, such as serpentines. We present a study of D/H inter-diffusion in antigorite, a common variety of serpentine. High-temperature (HT) experiments were performed in a belt apparatus at 540°C and 3.0 GPa on natural antigorite powders saturated with interstitial D2O. A low-temperature (LT) experiment was performed in diamond anvil cell at 350°C and 2.5 GPa on an antigorite single-crystal loaded with pure D<sub>2</sub>O. D/(D+H) ratios were mapped using Raman spectroscopy for the HT experiment and NanoSIMS for the LT experiment. As antigorite is a phyllosilicate, diffusion coefficients were obtained for crystallographic directions parallel and perpendicular to the antigorite layers (perpendicular and parallel to the c\*-axis, respectively). The equations of D/H inter-diffusion coefficients were determined to be  $D_{D/H}~(m^2/s) = 5.04~x~10^{-5}~x~exp(-170(\pm 53)~(kJ/mol)~/~RT)$  and  $D_{D/H}~(m^2/s) = 1.52~x~10^{-7}~x~exp(-157(\pm 32)~(kJ/mol)~/~RT)$  perpendicular and along the c\*-axis, respectively, and  $D_{D/H}~(m^2/s) = 7.29~x~10^{-6}~x~exp(-166(\pm 14))$ (kJ/mol) / RT) for the bulk diffusivity. These results are similar to those obtained on chlorite, in agreement with the similar crystallographic structures and atomic bonds in the two minerals. Assuming D/H inter-diffusion coefficients for antigorite are the same for all serpentine species, closure temperature and diffusion durations are applied to hydrothermal fields and in CI, CM and CR chondrites. Closure temperatures lie below 300°C for terrestrial hydrothermal alteration and depend on serpentine variety because they have different typical grain sizes. They lie below 130°C for carbonaceous chondrites, indicating that D/H isotopic exchange may have persisted down to very low temperatures on their parent bodies. D/H isotopic composition may be associated with grain size heterogeneities in carbonaceous chondrites due to protracted alteration of fine-grained material with the lowest closure temperatures (ca 50°C).