



## **Urban Climate Effects on Air Pollution and Atmospheric Chemistry**

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Tropospheric ozone, adversely affects the environment and human health. The presence of chlorine nitrate ( $\text{ClNO}_2$ ) in the troposphere can enhance ozone ( $\text{O}_3$ ) formation as it undergoes photolysis, releasing chlorine reactive atoms ( $\text{Cl}$ ) and nitrogen dioxide ( $\text{NO}_2$ ), both of which enhance tropospheric ozone formation. The importance of new sources of tropospheric  $\text{ClNO}_2$  via heterogeneous processes has recently been highlighted. This study employed a box model, using the Master Chemical Mechanism (MCM version 3.2) to assess the effect of  $\text{ClNO}_2$  on air quality in urban areas within the UK. The model updated to include  $\text{ClNO}_2$  production, photolysis, a comprehensive parameterisation of dinitrogen pentoxide ( $\text{N}_2\text{O}_5$ ) uptake, and  $\text{ClNO}_2$  production calculated from bulk aerosol composition. The model simulation revealed the presence of  $\text{ClNO}_2$  enhances the formation of  $\text{NO}_2$ , organic peroxy radical ( $\text{CH}_3\text{O}_2$ ),  $\text{O}_3$ , and hydroxyl radicals ( $\text{OH}$ ) when compared with simulations excluding  $\text{ClNO}_2$ . In addition, the study examined the effect of temperature variation upon  $\text{ClNO}_2$  formation. The response of  $\text{ClNO}_2$  to temperature was analysed to identify the underlying drivers, of particular importance when assessing the response of atmospheric chemistry processes under potential future climates.