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### **Water and anion transport conversion in highly heterogeneous, recultivated open mining fields with very different carbon levels and pH values: multitracer lysimeter studies**

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The movement of water and the related transport of bromide and nitrate were studied in soil used for agriculture from an open mining field near Espenhain/Saxony by means of tracer techniques (D<sub>2</sub>O, [<sup>15</sup>N]nitrate, Br tracer) in monolithic and reconstructed lysimeters. Despite the similarly high level of seepage, the breakthrough volumes were very different, namely 82 l (L 12/1), 147 l (L 12/3) and 30 l (L124) for D<sub>2</sub>O. After the experiments had continued for 36 months, the recovery standardised for 600 l seepage of bromide and deuterium respectively were calculated to be 22% and 39% for L 12/1, 15% and 19% for L 12/3, and 4% and 46% for L 124. The differences in the seepage recovery of the reactive tracer [<sup>15</sup>N] nitrate were even greater for the three lysimeters, namely 3,7% (L 12/1), 0,7 % (L 12/3) and 1,5 % (L 124).

The findings regarding the transport of D<sub>2</sub>O indicate that water transport in the soil monolith of L 12/1 is mainly determined by preferential flow, whereas in L 12/3 and L 124 conditions are largely shaped by piston flow and delays corresponding to the cascade model. Taking into account plant uptake, the sometimes much lower recovery of bromide compared to D<sub>2</sub>O in lysimeter L 12/1 and especially in L 124 could be attributed to reactions resulting from the strong acidic conditions (pH 1.5–2.0) in the soil water of these lysimeters, the extremely high sulphate levels (over 4%) and/or the reactive carbon from lignite residues. Under these conditions, the bromide appears to undergo temporary chemisorption and possibly even chemical conversion. The seepage recovery of [<sup>15</sup>N] nitrate in the open mining field soil is significantly lower than in natural soils. [<sup>15</sup>N] nitrate recovery in 600 l seepage of 3,7% (L 12/1), 0,7% (L 12/3) and 1,5% (L 124) are probably caused by nitrate decomposition via denitrification under the partly extreme soil conditions.