## TAILORING OF A HIERARCHICALLY STRUCTURED MATERIAL FROM DIATOMITE

Höllen, D.<sup>1</sup>, Klammer, D.<sup>2</sup>, Letofsky-Papst, I.<sup>3</sup> & Dietzel, M.<sup>2</sup>

<sup>1</sup>Montanuniversität Leoben, Chair of Waste Processing Technology and Waste Management Erzherzog-Johann-Str. 18, 8700 Leoben, Austria <sup>2</sup>Graz University of Technology, Institute of Applied Geosciences Rechbauerstraße 12, 8010 Graz, Austria <sup>3</sup>Austrian Centre for Electron Microscopy and Nanoanalytics Steyrergasse 17, 8010 Graz, Austria e-mail: daniel.hoellen@unileoben.ac.at

Based on a previous study (HÖLLEN et al., 2012), where diatomite was converted into zeolites via the formation of an intermediate phase we adapted the hydrothermal treatment process by decreasing the molarity of the Al-containing KOH solution from 1 M to 0.1 M to slow down the dissolution of diatoms and to prolong the period of metastability of the intermediate phase. This change of experimental parameters yielded after 1 d at 100°C a hierarchically structured material consisting of remaining diatoms with macropores of about 100 nm and newly formed nanoparticles of the intermediate phase. These x-ray amorphous particles consist of a potassium-aluminium-hydroxy-silicate, have a diameter of about 50 nm and are characterized by inner pores with a diameter of only few nm. The intermediate phase can remove heavy metal ions like  $Cu^{2+}$ ,  $Pb^{2+}$  and  $Zn^{2+}$  very efficiently from aqueous solution. Considering that particulated matter acts as adsorbens for oxyanionic contaminants like  $AsO_4^{3-}$ ,  $CrO_4^{2-}$  and  $MoO_4^{2-}$  which can be formed during leaching of alkaline wastes (CORNELIS et al., 2008), hierarchically structured materials which can remove dissolved and particulate contaminants simultaneously from aqueous solutions are highly promising.

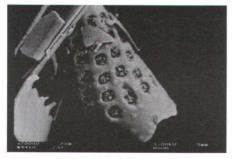


Figure 1. SEM-SE image of a hierarchically structured material, consisting of a macroporous diatom and nanoporous potassium-aluminium-hydroxosilicates growing in its pores.

CORNELIS, G., JOHNSON, C.A., VAN GERVEN, T VANDESCASTEELE, C. (2008): Applied Geochemistry, 23, 5, 955-976.

HÖLLEN, D., KLAMMER, D.; LETOFSKY-PAPST, I, DIETZEL, M. (2012): Journal of Material Science and Engineering A & B., 2, 10, 523-533.