

COMPOSITIONAL VARIATION OF Ba-RICH WHITE MICAS FROM TWO DIFFERENT GEOLOGICAL SETTINGS

Tribus, M.¹, Pomella, H.², Tropper, P.¹ & Linner, M.³

¹Institute of Mineralogy and Petrography, Faculty of Geo- and Atmospheric Sciences, University of Innsbruck, Innrain 52, 6020 Innsbruck, Austria

²Institute of Geology and Paleontology, Faculty of Geo- and Atmospheric Sciences, University of Innsbruck, Innrain 52, 6020 Innsbruck, Austria

³Geological Survey of Austria, Neulinggasse 38, 1030 Wien, Austria
e-mail: martina.tribus@uibk.ac.at

Ba-micas can occur in a variety of geological environments. The formation of Ba-rich micas is strongly influenced by the presence of hydrothermal fluids and thus can be classified as a product of metasomatic processes. Ba-rich muscovite/paragonite, so called ganterite, is a dioctahedral white mica represented by a 1:1 mixture between the true micas muscovite/paragonite and the hypothetical Ba brittle mica with an ideal composition of $[\text{Ba}_{0.5}(\text{Na}+\text{K})_{0.5}]\text{Al}_2[\text{Si}_{2.5}\text{Al}_{1.5}]\text{O}_{10}(\text{OH})_2$ (GRAESER et al., 2003; JAMBOR & ROBERTS, 2004). As the interlayer position is filled up to 50% by monovalent cations ganterite is a member of the true micas.

(1) The Ba-rich white micas of the first study occur in the Lienz contact aureole adjacent to the Oligocene Lienz/Edenwald tonalite in the surrounding fine-grained mica schists of the Austroalpine basement. Thermometric calculations for the innermost part of the aureole yielded temperatures of 640 ± 24 °C. The Ba content in the white mica reaches up to 13.38 wt.% BaO, even though Ba is not the dominant interlayer cation. The formula of a typical Ba-rich white mica is $[\text{Ba}_{0.37}\text{K}_{0.41}\text{Na}_{0.22}]_{1.00}[\text{Al}_{1.93}\text{Mg}_{0.02}\text{Fe}_{0.03}\text{Ti}_{0.02}]_{2.00}[\text{Si}_{2.66}\text{Al}_{1.34}]\text{O}_{10}(\text{OH})_2$. Based on the chemical composition of the mica endmembers paragonite and muscovite, the Ba-rich mica in this study can be formed by a combination of the coupled substitution $[\text{Ba}][\text{Al}^{\text{IV}}]=[\text{K}]_{-1}[\text{Si}]_{-1}$ and the simple $[\text{Na}]=[\text{K}]_{-1}$ exchange vector. Complete solid solutions between muscovite and Ba-rich white micas were observed since Ba contents range from 0.07 wt.% up to 13.38 wt.% BaO.

(2) The Ba-rich white mica of the second study was found in the highly deformed marble belt close to the Meran-Mauls fault, which is part of the Giudicarie fault system. The marble layer occurs within the paragneisses of the Meran-Mauls Basement and has been overprinted under Variscan amphibolite-facies and eo-Alpine/Alpine greenschist-facies condition. The Ba-rich white mica is associated with celsian + barite + calcite + dolomite. The variation in the BaO content of the micas ranges from 0.49 wt.% up to 8.78 wt.%. As most of the white micas in this study are “phengitic”, the incorporation of Ba can be described by the coupled substitution $[\text{Ba}][\text{Al}^{\text{VI}}][\text{Al}^{\text{IV}}]_2 = [\text{K}]_{-1}[\text{Mg}]_{-1}[\text{Si}]_{-2}$. The formula of a low Ba-mica (0.96 wt.% BaO) is $\square_{0.10}[\text{Ba}_{0.02}\text{K}_{0.86}\text{Na}_{0.01}\text{Ca}_{0.01}]_{0.90}[\text{Al}_{1.56}\text{Mg}_{0.42}\text{Fe}_{0.01}\text{Ti}_{0.01}]_{2.00}[\text{Si}_{3.48}\text{Al}_{0.52}]_{4.00}\text{O}_{10}(\text{OH})_2$. This indicates a 50:50 composition along the muscovite – aluminoceladonite solid solution.

GRAESER, S., HETHERINGTON, C., GIERÉ, R. (2003): *Canadian Mineralogist*, 41, 1271-1280.
JAMBOR, J.L., ROBERTS, C. (2004): *American Mineralogist*, 89, 1826-1834.