

## STRUCTURAL CHARACTERIZATION OF LANNONITE FROM THE ANNA MINE, ALSDORF, GERMANY

Oberwandler, L.<sup>1</sup>, Pristacz, H.<sup>1</sup>, Kolitsch, U.<sup>2,1</sup> & Lengauer, C.L.<sup>1</sup>

<sup>1</sup> Institut für Mineralogie und Kristallographie, Universität Wien – Geozentrum,  
Althanstrasse 14, 1090 Wien, Austria

<sup>2</sup> Mineralogisch-Petrographische Abteilung, Naturhistorisches Museum Wien  
Burgring 7, 1010 Wien, Austria.  
e-mail: christian.lengauer@univie.ac.at

Lannonite and wilcoxite were first described by WILLIAMS & CESBRON (1983) as two new fluosulfates from the Lone Pine mine, New Mexico, where both occur as weathering products of the primary pyrite-rich ore. The chemical formula was reported to be  $\text{HMg}_2\text{Ca}_4\text{Al}_4(\text{SO}_4)_8\text{F}_9 \cdot 32\text{H}_2\text{O}$  and a tetragonal symmetry without centering was derived from X-ray powder data. Due to the quality of the type material, however, no structural information could be obtained for both minerals. Recently PETERSON & JOY (2013) presented a detailed structural description of wilcoxite,  $\text{MgAl}(\text{SO}_4)_2\text{F} \cdot 17\text{H}_2\text{O}$ .

A second finding of lannonite from the Anna Mine, Alsdorf near Aachen, Germany, was reported by BLASS & STREHLER (1993), where lannonite is formed during the burning and weathering process of a coal dump after spontaneous ignition, i.e. mobilization of volatile components from the coal and formation of acidic solutions, which decompose the surrounding rock. Beside several ammonium and sulfate minerals, e.g. ammoniojarosite, anhydrite, or thermessaite-(NH<sub>4</sub>), selenium is also observed in the lannonite-bearing paragenesis. The title compound occurs as clear, colourless, optically uniaxial, tetragonal (square) platelets, which are suitable for single-crystal X-ray investigations at ambient temperature. Even though lannonite has a reported H<sub>2</sub>O content of 32 wt.%, it is a stable mineral with Mohs hardness of about 2 and a reported density of 2.22 g/cm<sup>3</sup>.

The extinction conditions revealed a tetragonal *I*-centered cell ( $a = 6.860(1)$ ,  $c = 28.053(5)$  Å,  $V = 1320.3(4)$  Å<sup>3</sup>), and consecutive structure refinements applying the space groups of Laue class *4/m* revealed the correct space group to be *I4/m* ( $R_{\text{1all}} = 4.25\%$ ). The first outcome of the refined structure model is a reduced H<sub>2</sub>O and F content leading to  $\text{Mg}_2\text{Ca}_4\text{Al}_4(\text{SO}_4)_8\text{F}_8 \cdot 24\text{H}_2\text{O}$  as the corrected chemical formula for lannonite ( $D_x$  of 2.100 g/cm<sup>3</sup>) from this locality. The structure can be described by a columnar sequence along [001] of F-linked CaFO<sub>5</sub>-AlF<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub>-MgF<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub>-AlF<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub>-CaFO<sub>5</sub> octahedra terminated by positionally disordered SO<sub>4</sub> tetrahedra. These columns are interlinked by a second type of SO<sub>4</sub> tetrahedra, connecting neighbouring CaFO<sub>5</sub> octahedra.

The crystal-structure determination, together with optical and chemical-analytical data and Raman spectroscopy of the material will be presented and discussed.

We thank Mr. Frank de Wit for kindly providing the samples studied.

BLASS, G., STREHLER, H. (1993): Mineralien-Welt, 4(4), 35-42.

PETERSON, R.C., JOY, B.R. (2013): Canad. Mineral., 51, 107-117.

WILLIAMS, S.A., CESBRON, F.P. (1983): Mineral. Mag., 47, 37-40.