THE CRYSTAL STRUCTURE OF TYROLITE, Ca₂Cu₉(AsO₄)₄(CO₃)(OH)₈(H₂O)₁₀: SYNCHROTRON X-RAY DIFFRACTION STUDY

Krivovichev, S. V. ^{1,2}, Chernyshov, D. Yu. ³, Döbelin, N. ⁴, Kahlenberg, V. ², Tessadri, R. ², Armbruster, T. ⁴ & Kaltenhauser, G. ²

¹ Department of Crystallography, Faculty of Geology, St.Petersburg State University, St.Petersburg 199034, Russia

² Institut für Mineralogie und Petrographie, Universität Innsbruck, Innrain 52, A-6020 Innsbruck, Austria

³ Swiss-Norwegian beamline, European Synchrotron Radiation Facility, BP 220, Grenoble 38043, France

⁴ Laboratorium für chemische und mineralogische Kristallographie, Universität Bern,

Freiestrasse 3, CH-3102 Bern, Switzerland

e-mail: skrivovi@mail.ru

Tyrolite, Ca₂Cu₉(AsO₄)₄(CO₃)(OH)₈(H₂O)₁₀, is common mineral from oxidation zones of sulfide mineral deposits. First described from Schwaz-Brixlegg locality in Tyrol, Austria, it has been subsequently reported for about 128 localities all over the world. Tyrolite was described as orthorhombic, a = 10.212, b = 55.510, c = 5.602 Å, V = 3175.6 Å³ [PDF # 11-348]. In 1980, a monoclinic tyrolite-like mineral was reported by MA et al. (1980) from Dongchuan copper mine, Yunnan Province, China. The mineral had the same chemistry as tyrolite but was identified as monoclinic, a = 10.513, b = 5.560, c = 27.610 Å, = 94, V = 1609.9 Å³ The mineral was named 'clinotyrolite' but it has not been approved by the CNMMN IMA. For this study, a crystal from the type locality (Schwaz in Tyrol) was selected. X-ray diffraction experiments were performed at the Swiss-Norwegian beamline BM01 of the ESRF with an imaging plate area detector Mar345. We have found tyrolite to be monoclinic, P2/c, a = 27.562(3), b = 5.5682(7), c = 10.466(2) Å, $= 98.074(11)^{\circ}$, V = 1590.3(3) Å³ There are five Cu(II) sites in Jahn-Teller-distorted octahedral coordinations to

six anions. The structure (Figure) is based upon 2nm-thick copper arsenate layers with complex architecture. The layers are insulated by Ca^{2+} cations, CO_3^{2-} groups and H₂O molecules. The layers are held together by hydrogen bonds only.

References

[I] MA, Z., QIAN, R.Y., PENG,
Z.-Z. (1980): Dizhi Xuebao, 54,
134-143.

