

RAMAN SPECTRA OF TITANOSILICATE MELTS

Korinevskaja, G. & Bykov, V.

Institute of Mineralogy, Miass, 456317, Russia
e-mail: kor@ilmeny.ac.ru

Titanium is a minor element in most rock-forming silicate melts. Nevertheless, this element is petrologically significant (MYSEN et al., 1980). Titanium is in six-fold coordination by oxygen in the majority of natural silicates, and titanium can replace silicon in four-fold coordination in alkaline minerals. A structural feature of titanosilicate glasses is that titanium can be both in six-fold coordination, and in four-fold coordination. Structure of melts of systems: 33%Na₂O·67%SiO₂-x%TiO₂, 40%Na₂O·60%SiO₂-x%TiO₂, 50%Na₂O·50%SiO₂-x%TiO₂ (x=1, 5, 10, 20 %) have been investigated by Raman high-temperature spectroscopy. The pulse laser was used for excitation of spectra and a synchronized system of account of photons was used for the registration. It is necessary to discriminate a thermal background from the heating furnace and melt.

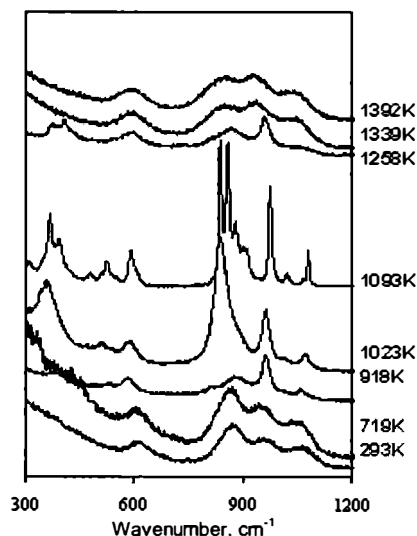


Fig. 1. Raman spectra of the compound with 50%Na₂O-45%SiO₂-5%TiO₂ at different temperatures

As an example, Raman spectra of the compound with 50%Na₂O-45%SiO₂-5%TiO₂ are shown in Fig. 1 at temperatures of 293-1389 K. Bands in the range of 1050-1075 cm⁻¹, 940-970 cm⁻¹, 860-880 cm⁻¹ and 570-625 cm⁻¹ are observed in the spectra of titanosilicate glasses and melts. Bands at 1050-1075 cm⁻¹ and 940-970 cm⁻¹ are attributed to the highly localized symmetric Si-nonbridging O stretching vibrations of Q³ and Q² species, respectively (MYSEN & NEUVILLE, 1995). The low frequency bands are associated with stretching vibrations of Si-O-Si linkages. The band at 860-880 cm⁻¹ is assigned to vibrations of TiO₄ units (FURUKAWA & WHITE, 1979). Raman spectra of crystalline phases which are formed at heat treatment of glasses at temperatures of 800-1000 K were obtained too. Systematics of the band behavior in Raman spectra depending on composition (SiO₂ and TiO₂ content) and temperature are established.

References

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