

TEMPERATURE DEPENDENCE OF THE Fe^{2+} MÖSSBAUER PARAMETERS IN TRIPHYLITE (LiFePO_4)

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Triphylite (LiFePO_4) has an olivine type structure (Pnma) and has received much attention because of its application as possible electrode active material for lithium rechargeable batteries.

In this work, the temperature dependence of the Mössbauer parameters of Fe^{2+} in synthetic triphylite (ARNOLD et al., 2003) is determined between 4 K and room temperature. From the Mössbauer measurements a sharp magnetic ordering transition is observed at 53 K, which is in line with the Néel temperature as is determined by neutron diffraction (ROUSSE, 2003). The Mössbauer spectra in the paramagnetic region are well fitted by one Fe^{2+} quadrupole doublet with small linewidth. The spectra in the magnetic region are typical for Fe^{2+} , showing the maximum of eight absorption lines. They are analysed using the diagonalization of the full nuclear-interaction Hamiltonian and could be successfully fitted with a model independent distribution of hyperfine fields.

The hyperfine field is always collinear to the local principal axis of the electric field gradient tensor. The asymmetry parameter is about 0.79 and the quadrupole coupling constant is 2.79 mm/s, which is perfect in line with the quadrupole splitting $\Delta E_Q = 3.06$ mm/s at 80 K in the paramagnetic region.

The temperature dependence of the quadrupole splitting is interpreted within the ^5D orbital energy level scheme of Fe^{2+} by a crystal field calculation (VAN ALBOOM et al., 1993) based on the point symmetry of the Fe^{2+} site in triphylite (ANDERSON et al., 2000).

When triphylite is used as electrode material in lithium rechargeable batteries, there is a conversion of triphylite in its delithiated form FePO_4 (heterosite) and vice versa during the charge/discharge process. The characteristic Mössbauer temperature is an important parameter in order to determine accurate ratios of both compounds during the process (ANDERSON et al., 2000). From the temperature dependence of the isomer shift its value is determined to 390 K for Fe^{2+} in triphylite. The intrinsic isomershift is 1.45 mm/s.

References

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